



19 **Abstract**

20 Tetracycline antibiotics (TA) used in veterinary medicine reach terrestrial ecosystems  
21 mostly via the repeated applications of animal manures and slurries on agricultural soils,  
22 where they may cause toxic effects on bacterial communities. In the current work, we  
23 studied the efficacy of adding doses of 0, 6, 24 and 48 g kg<sup>-1</sup> of biomass ash (BA) to  
24 four different soils to reduce potential negative effects of tetracycline antibiotics.  
25 Specifically, soil samples were polluted with different concentrations of tetracycline,  
26 oxytetracycline or chlortetracycline, and the bacterial community growth was estimated  
27 using the <sup>3</sup>H leucine incorporation technique. Soil amendment with BA increased soil  
28 pH (1.3-4.8 units), total carbon (0.7-5.8 g kg<sup>-1</sup>) and Fe and Al oxides concentrations  
29 (0.25-3.98 g kg<sup>-1</sup>), as well as bacterial activity (1-9 times compared to the control). In  
30 addition, BA amendment at high doses (24 or 48 g kg<sup>-1</sup>) resulted in a similar toxicity  
31 decrease for the three antibiotics, but with variations among soils. The reductions in  
32 antibiotics toxicity were very variable, ranging between 5% and 100% (total recovery).  
33 In view of that, the spreading of BA could be interesting as management practice to  
34 reduce risks of soil pollution and subsequent toxicity on bacterial communities due to  
35 tetracycline antibiotics.

36

37 **Keywords:** Biomass ash; chlortetracycline; leucine incorporation; oxytetracycline;  
38 tetracycline

39

40 **1. Introduction**

41 Veterinary antibiotics are widely used for animal health, and even as animal  
42 growth promoters in countries where this parallel use is not banned (Anderson et al.,  
43 2005). These antibiotics reach terrestrial ecosystems via the repeated applications of

44 animal manures and slurries as amendments on agricultural soils (Pan and Chu, 2017).  
45 Once in the soil, veterinary antibiotics may cause harmful effects on non-target  
46 organisms, such as soil bacterial communities (Warman, 1980). The presence of  
47 antibiotics in the soil may cause a reduction in microbial biodiversity (Thiele-Bruhn and  
48 Beck, 2005; Hammesfahr et al., 2008), influence the growth and enzymatic activities of  
49 bacterial communities (Liu et al., 2009; Ma et al., 2016; Song et al., 2017; Santás-  
50 Miguel et al., 2020), and hence, ecological functions and functional stability (Zielezny  
51 et al., 2006; Demoling and Bååth, 2008; Pallecchi et al., 2008; Toth et al., 2011).

52 Tetracyclines are the antibiotics most used in the European Union, especially  
53 tetracycline, oxytetracycline and chlortetracycline (ESVAC, 2016), which motivated  
54 that previous studies have focused on the effects of tetracycline residues on soil  
55 microbial functions (Thiele-Bruhn, 2005; Yang et al., 2010; Liu et al., 2012; Ma et al.,  
56 2016; Song et al., 2017). It has been established that the main agents responsible for the  
57 sustainability of soils are microorganisms (Pulleman et al., 2012), since they are  
58 involved in nutrient cycles and renewal of organic matter (Thiele-Bruhn et al., 2012).  
59 Therefore, the correct growth of soil bacterial communities will help to maintain an  
60 optimal soil fertility and organic matter turnover. However, there are only few studies  
61 estimating the effects of tetracycline antibiotics on soil bacterial community growth  
62 (Rousk et al., 2008; Rousk et al., 2009a; Santás-Miguel et al., 2020). These studies have  
63 shown a marked toxicity due to tetracycline antibiotics, which was highly dependent on  
64 soil characteristics.

65 In order to prevent harmful effects of antibiotics accumulated into soils, the use  
66 of appropriate waste and/or by-products as soil amendments may be a low cost  
67 alternative aiding to decrease undesirable toxicity affecting soil bacterial communities.  
68 It has been previously demonstrated that soil amendment with certain by-products, such

69 as mussel-shell (Fernández-Pazos et al., 2013; Osorio-López et al., 2014; Otero et al.,  
70 2015; Romar-Gasalla et al., 2018), pyritic material (Fernández-Pazos et al., 2013;  
71 Osorio-López et al., 2014; Otero et al., 2015), pine bark (Romar-Gasalla et al., 2018),  
72 oak ash (Romar-Gasalla et al., 2018) and hemp waste (Romar-Gasalla et al., 2018) may  
73 reduce the availability of different types of pollutants, both anionic and cationic.  
74 However, even if is considered that the amendment of soils with by-products may  
75 stimulate or inhibit microbial communities, the potential effects of by-products on soil  
76 bacterial community growth has been poorly studied. Within the few researches  
77 previously carried out in this regard, Fernández-Calviño et al. (2015a) assessed the  
78 effect of different doses of crushed mussel shell applied in two Cu-polluted acid mine  
79 soils, showing long-term positive effects. In a subsequent study, Fernández-Calviño et  
80 al. (2018) measured bacterial and fungal growth in acid mine soils amended with pine  
81 bark, crushed mussel shell and a mixture of both, showing general positive effects on  
82 microbial growth. However, no previous studies have dealt with the effects of biomass  
83 ash amendment on bacterial community growth in soils.

84 Biomass ash (mainly derived from biomass energy plants and combustion  
85 boilers of variate sizes) is a very abundant waste material, which increases soil pH, as  
86 well as the contents of major nutrients, and decreases the availability of Al and some  
87 minor elements (Demeyer et al., 2001). Also, biomass ash may reduce the availability of  
88 different pollutants in soils (Núñez-Delgado et al., 2015; Núñez-Delgado et al., 2017;  
89 Romar-Gasalla et al., 2018). Specifically regarding effects on microbiota, Zimmermann  
90 and Frey (2002) studied the effect of wood ash amendment on soil microbial process,  
91 showing an increase in pH and a subsequent stimulation of microbial biomass, basal  
92 respiration and enzymatic activities.

93 In view of this background, in the current work we hypothesize that adding  
94 appropriate doses of biomass ash to soils, the bacterial community growth would be  
95 stimulated, and toxicity effects due to tetracycline antibiotics would decrease.  
96 Therefore, the main objective of this work is to assess the effect of biomass ash  
97 amendments on bacterial community growth in soils polluted with the tetracycline  
98 antibiotics tetracycline, oxytetracycline and chlortetracycline.

99

## 100 **2. Material and Methods**

### 101 **2.1. Chemicals**

102 Tetracycline hydrochloride (CAS. 64-75-5;  $\geq 95\%$  in purity), Oxytetracycline  
103 hydrochloride (CAS 2058-46-0;  $\geq 95\%$  in purity) and Chlortetracycline hydrochloride  
104 (CAS 64-72-2;  $\geq 97\%$  in purity), all three supplied by Sigma–Aldrich (Steinheim,  
105 Germany), were used for soil spiking.

106

### 107 **2.2. Procurement of soil and biomass ash samples**

108 Four soils devoted to cultivation of potatoes in rotation with cereal, located in  
109 Galicia (NW of Spain), previously analyzed by Conde-Cid et al. (2019a), were selected  
110 for the current study. On each sampling site 10-20 sub-samples were taken in the soil  
111 surface horizon (0-20 cm) using an Edelman probe, and were subsequently mixed in  
112 one composite sample. At the laboratory, the soil samples were air dried, sieved by a 2  
113 mm mesh light screen, homogenized and stored in polyethylene bottles. Soil pH was  
114 determined in water (soil ratio: 1:2.5), using a combined glass electrode. Total carbon  
115 and total nitrogen were determined by elemental analysis in a LECO CHN-1000 (LECO  
116 Corporation, St. Joseph, MI, USA). Exchangeable basic cations (Ca, Mg, Na and K)  
117 were extracted with 0.2 M  $\text{NH}_4\text{Cl}$  (Sumner and Miller, 1996), while exchangeable Al

118 was extracted with 1 M KCl (Bertsch and Bloom, 1996), and then determined by flame  
119 atomic absorption (Ca, Mg and Al) or emission spectroscopy (Na and K). The effective  
120 cation exchange capacity (eCEC) was estimated as the sum of the exchangeable basic  
121 cations and Al. Fe and Al oxides ( $Fe_o$  and  $Al_o$ ) were extracted with 0.2 M ammonium  
122 oxalate-oxalic acid (Blakemore, 1978).

123 The main characteristics of these soils are shown in Table 1. Briefly, these soils  
124 present pH values (measured in water) between 4.6 and 5.1, organic carbon contents  
125 between 1.1 and 3.8%, and effective cation exchange capacity between 4.1 and 6.8  
126  $cmol_c\ kg^{-1}$ . Al and Fe oxides have values ranging between 855-5040  $mg\ kg^{-1}$ , and  
127 between 1150-2585  $mg\ kg^{-1}$ , respectively.

128 The biomass ash (BA) used was from a combustion boiler in Lugo (Spain).  
129 Previously, this by-product was analyzed by Romar-Gasalla et al. (2018). Briefly, it  
130 showed a pH in water of 11.3, a total carbon content of 11.7%, a total nitrogen content  
131 of 0.21%, and a C/N ratio of 55.5. It presented high contents of amorphous Al and Fe  
132 oxides (8323 and 4233  $mg\ kg^{-1}$  respectively), and also relatively high levels of oxides of  
133 Ca and other elements (Table 1), due the combustion process.

134

### 135 **2.3. Experimental Design**

136 Four soils were amended with four doses of biomass ash (0, 6, 24 and 48  $g\ kg^{-1}$ ),  
137 resulting in 16 soil mixtures. In brief, 288 dry grams of each soil sample were weighed  
138 in four 500 mL polyethylene bottles (72 g per bottle) and mixed with different amounts  
139 of biomass ash (0, 0.43, 1.73, 3.46 g) in order to obtain the final concentrations of 0, 6,  
140 24 and 48 g of biomass ash per  $kg^{-1}$  of soil. These concentrations were satisfactorily  
141 used in previous works testing the effects of different by-products on soils (Ramírez-  
142 Pérez et al., 2013; Fernández-Calviño et al., 2015a; Conde-Cid et al., 2020a; Conde-Cid

143 et al., 2020b). Later on, the resulting mixtures of dry soil and different concentrations of  
144 biomass ash were rewetted up to 60-80% of water-holding capacity, and the bottles  
145 sealed with lab-film (Parafilm) to avoid soil drying. Then, these mixtures were  
146 incubated during 30 days at 22 °C in the dark.

147 After this incubation time, different amounts of tetracycline, oxytetracycline and  
148 chlortetracycline were dry added (via talc) separately (by triplicate) to the 16 soil  
149 mixtures in 50 mL polypropylene tubes, reaching 8 concentrations for each antibiotic  
150 and each mixture (0, 0.5, 2, 7.8, 31.3, 125, 500 and 2000 mg kg<sup>-1</sup>). The bottles were  
151 sealed with Parafilm to avoid soil drying (thus maintaining water-holding capacity). The  
152 resulting microcosms (totalizing a number of 960, 1 g each one) were initially studied  
153 for 1 day, focusing on the assessment of the subsequent bacterial community growth,  
154 performed after extraction from soil samples, and estimated using the <sup>3</sup>H Leucine  
155 incorporation method (Bååth 1994; Bååth et al., 2001). Briefly, 1 g of soil (fresh  
156 weight) was mixed with 10 mL distilled water using a multivortex shaker for 3 min at  
157 maximum intensity, followed by low-speed centrifugation at 1000 x g for 10 min, to  
158 create a bacterial suspension in the supernatant. An aliquot (1 mL) of this suspension  
159 was transferred to 2 mL micro-centrifugation tubes. Then, 2 µL [<sup>3</sup>H]Leu (3.7 MBq mL<sup>-1</sup>  
160 and 0.574 TBq mmol<sup>-1</sup>; Perkin Elmer, USA) were added with non-labeled Leu to each  
161 tube, resulting in 275 nM Leu in the bacterial suspensions. After incubation for 2 h at 22  
162 °C, bacterial growth was stopped with 75 µL 100% trichloroacetic acid. Washing was  
163 performed as described by Bååth et al. (2001), and the amount of leucine incorporated  
164 was determined using scintillation liquid counting (Tri-Carb 2810 TR, Perkin Elmer,  
165 USA).

166

#### 167 **2.4. Data analysis**

168 Data on bacterial growth obtained for each soil and each concentration of  
169 tetracycline, oxytetracycline or chlortetracycline were normalized (Santás-Miguel et al.,  
170 2020) respect to the control (samples without antibiotic) to be able to compare the  
171 inhibition curves of different soils and concentrations of biomass ash. The normalized  
172 data were represented as inhibition curves of relative bacterial growth as a function of  
173 antibiotic doses. From the inhibition curves, a toxicity index ( $\log IC_{50}$ ) was calculated to  
174 compare the effects of different doses of BA regarding the toxicity exerted by  
175 antibiotics on soil bacterial communities. The logarithm of the concentration that  
176 inhibits 50% of bacterial growth ( $\log IC_{50}$ ) was calculated using the following logistic  
177 model:  $Y = c/[1 + e^{b(a-x)}]$ , where  $Y$  is Leu incorporation for each antibiotic concentration,  
178  $x$  is the logarithm of the concentration of antibiotic added,  $a$  is the value of  $\log IC_{50}$ ,  $b$  is  
179 a parameter related to the slope of inhibition curves, and  $c$  is the bacterial growth  
180 without antibiotic addition.

181

### 182 **3. Results and discussion**

#### 183 *3.1. Biomass ash effect on general soil characteristics and bacterial community growth*

184 In soils without BA (dose 0 g kg<sup>-1</sup>) the values of soil pH varied between 4.6 and  
185 5.1 (Table 1). However, soil pH changed significantly after biomass ash addition,  
186 following dose-response curves (Fig. 1). Thus, in soil samples amendment with 6 g kg<sup>-1</sup>  
187 of biomass ash, pH increased between 1.3 and 2.4 units. Regarding soils amended with  
188 24 g kg<sup>-1</sup> and 48 g kg<sup>-1</sup> of BA, pH values increases between 2.9 and 4.2 units, and  
189 between 3.6 and 4.8 units, respectively. Total carbon in soils without BA varied  
190 between 11 and 38 g kg<sup>-1</sup> (Table 1), values in the range of those usually found in Galicia  
191 (Calvo de Anta et al., 2015). Carrying out a theoretical calculation, the addition of BA  
192 (which contains 11.7% of C, Table 1) to agricultural soils contributed to an increase of

193 total carbon amounting: 0.7 g kg<sup>-1</sup> for a dose of 6 g kg<sup>-1</sup> of BA; 2.9 g kg<sup>-1</sup> for a dose of  
194 24 g kg<sup>-1</sup> of BA; and 5.8 g kg<sup>-1</sup> for a dose of 48 g kg<sup>-1</sup> of BA. The contents of Fe and Al  
195 oxides in the soils studied varied between 1.15 and 2.59 g kg<sup>-1</sup>, and between 0.86 and  
196 5.04 g kg<sup>-1</sup>, respectively (Fe<sub>o</sub> and Al<sub>o</sub> contents in Table 1). Carrying out the theoretical  
197 calculation of Fe and Al oxides added to each soil by BA amendment, the results  
198 indicated that the addition of 6 g kg<sup>-1</sup> of BA to soils contributed to an increase of 0.25  
199 and 0.50 g kg<sup>-1</sup> in the contents of Fe and Al oxides; while the increase ranged between  
200 1.01 and 1.99 g kg<sup>-1</sup> for a BA dose of 24 g kg<sup>-1</sup>; and reached up to between 2.02 and  
201 3.98 g kg<sup>-1</sup> for a dose of 48 g kg<sup>-1</sup> of BA (Table 2).

202 The behavior regarding mean bacterial growth in response to soil amendment with  
203 biomass ash (Fig. 2) was different for each soil. In soil 1 and 2 it was observed that the  
204 effect on bacterial growth due to BA addition was low (increases between 1-1.9 times as  
205 compared to control), while it was high for soils 3 and 4 (increases of up to 9.8 and 7.0  
206 times as compared to the control). Previous studies also found increases in bacterial  
207 community growth in response to soil amendment with various by-products, such as  
208 crushed mussel shells (Fernández-Caviño et al., 2015a), pine bark (Fernandez-Calviño  
209 et al., 2018), and marble wastes or organic wastes (Zornoza et al., 2016). These  
210 increases were generally attributed to both increases in soil pH and in total carbon, two  
211 characteristics also increased by biomass ash amendments. In relation to soil pH, Rousk  
212 et al. (2009b) found increases of 5-fold in bacterial community growth for a gradient of  
213 soil pH going from 4 up to 8. Organic matter also shows high importance for bacterial  
214 growth, and different authors found important increased in bacterial community growth  
215 in mine soils amended with organic materials (Zornoza et al., 2016; Fernández-Calviño  
216 et al., 2018).

217

218 3.2. *Effect of biomass ash on the toxicity exerted by tetracycline antibiotics on bacterial*  
219 *community growth*

220 Figure 3 shows the bacterial community growth inhibition curves as a function  
221 of the antibiotic dose, for the 3 antibiotics (tetracycline, oxytetracycline and  
222 chlortetracycline), as well as for the 4 soils studied and the 4 different BA doses added.  
223 Table 3 shows the estimated log IC<sub>50</sub> values obtained by fitting the inhibition curves to  
224 the logistic model. Log IC<sub>50</sub> values show that CTC was the most toxic antibiotic for  
225 bacterial community growth in absence of BA, while differences between OTC and TC  
226 are not clear. In the current study, adding BA to the soils, the toxicity exerted on  
227 bacterial community growth decreased for the three antibiotics (i.e. all three behave  
228 similarly). It was observed that, by increasing the biomass ash dose added to soils, the  
229 inhibition curves moved to the right with respect to the curve corresponding to the  
230 control (no BA added), and therefore the log IC<sub>50</sub> values increased. However, the BA  
231 dose needed to reduce significantly the toxicity of the antibiotics was different for the  
232 various soils here studied. Thus, in soils with low organic carbon (1.07%), the toxicity  
233 of the antibiotics was considerably reduced using 24 g kg<sup>-1</sup> of BA, with negligible  
234 inhibition of bacterial community growth being observed. Conversely, in the soils with  
235 higher organic carbon (3.1 to 3.8%), a clearly higher BA dose (48 g kg<sup>-1</sup>) was needed to  
236 decrease toxicity. These results suggest that the effectiveness of BA amendment for  
237 decreasing antibiotics toxicity on soil bacterial communities will be greater in soils with  
238 lower organic carbon than in those with higher organic carbon content.

239 The fact of achieving reductions of TC, OTC and CTC toxicity on bacterial community  
240 growth in presence of biomass ash may be due to different reasons. One reason may be  
241 that at high pH the tetracycline antibiotics became less toxic, since (according to Gesche  
242 et al., 1991) tetracyclines antimicrobial action is optimum at pH 6.0. Also, Gesche et al.

243 (2001) showed that oxytetracycline has a more effective antimicrobial action when it is  
244 on a substrate adjusted at acidic pH (around 6.0) than in substrates at basic pH (around  
245 8.0). The addition of BA to the soils used in the current study significantly increased the  
246 soil pH from acid to basic values, with magnitude dependent on BA dose. It is relevant  
247 that the effect of pH on the activity of the tetracycline antibiotics is due to their  
248 amphoteric character. Specifically, depending on soil pH, tetracycline antibiotics can be  
249 in cationic, zwitterionic or anionic form, and increases in pH due to biomass ash  
250 amendment may cause that the charge of tetracycline antibiotics go from cationic to  
251 zwitterionic or even anionic forms (Table 4). Furthermore, it must be noted that the  
252 species of these antibiotics determine their passage inside the bacteria, and therefore its  
253 toxicity. Tetracycline antibiotics diffuse through the plasmatic membrane of bacteria as  
254 positively charged cations (Yamaguchi et al., 1991), specifically through porin channels  
255 (Chopra et al., 1992), due the presence of Donnan equilibrium through the outer  
256 membrane (Stock et al., 1977). However, the diffusion of anionic tetracycline  
257 antibiotics species is hindered (Stock et al., 1977).

258 Another mechanism explaining reductions in the toxicity of tetracycline  
259 antibiotics in presence of biomass ash is the increase of antibiotics adsorption onto soils  
260 due to the amendment. In fact, biomass ash has a relevant capacity for retention of  
261 tetracycline antibiotics (Conde-Cid et al., 2019b). The increase of total carbon due to  
262 biomass ash amendment enhances the surface area in soils, facilitating antibiotics  
263 adsorption by soils, making them less bio-available and decreasing their toxicity on  
264 microbial communities. This hypothesis is consistent with results found by different  
265 authors who observed that organic carbon plays an important role in tetracyclines  
266 adsorption onto soils (Zhao et al., 2011a; Zhao et al., 2012; Fernández-Calviño et al.,  
267 2015a). In addition, Fernández-Calviño et al. (2015b) studied sorption kinetics for

268 tetracycline, oxytetracycline and chlortetracycline in two acid soils, with results  
269 showing high affinity for all three tetracyclines on both soils, although adsorption  
270 intensity was greater in the case of the soil containing more organic matter, clay and Al  
271 and Fe oxides. In the current work, the biomass ash amendment increased total soil  
272 carbon content, but also increased Al and Fe oxides ( $Fe_o$  and  $Al_o$ ), which may also  
273 contribute to tetracyclines adsorption onto soils. Besides, this by-product contains Ca,  
274 Fe (and other) oxides generated in the combustion process, and they may be related to  
275 sorption of anionic species through polyvalent cations, which can act as a cation bridge,  
276 linking organic matter and anionic species of the tetracycline antibiotics. In this sense,  
277 MacKay and Canterbury (2005) studied the adsorption of OTC on humic acids amended  
278 with metals in cationic form (Ca, Al and Fe), and observed that the addition of Al  
279 increased adsorption of OTC due to formation of ternary complexes between functional  
280 groups of the organic matter with negative charges, and zwitterion and anionic species  
281 of OTC. In addition, changes in soil pH can also affect to tetracycline antibiotics  
282 adsorption on soils, as shown by Figueroa and Mackay (2004), who investigated the  
283 adsorption of oxytetracycline on goethite and hematite and found that sorption increased  
284 with pH, reaching maximum values at about pH 8. Therefore, the effect of increasing  
285 soil pH due to the biomass ash amendment may also contribute to reduce tetracycline  
286 antibiotics toxicity on bacterial community growth.

287

#### 288 **4. Conclusions**

289 Soil amendment with biomass ash contributed to increase pH values, total  
290 carbon and Al and Fe oxides contents, increasing also the bacterial activity in the  
291 amended soils. Moreover, it can be a good alternative to reduce antibiotics toxicity  
292 exerted on bacterial community growth. However, the effectiveness of this practice will

293 be different depending on soil type. Specifically, in soils with low total carbon content  
294 the biomass ash doses needed for achieving clear reductions in antibiotics toxicity are  
295 lower than in soils with high total carbon contents. These results could be of relevance  
296 to program correct management practices for soils affected or in risk to be affected by  
297 pollution due to tetracycline antibiotics.

298

299

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483 **Table 1.** General characteristic of the soils and biomass ash studied. Average values  
 484 (n=3) with coefficients of variation always <5%

	<b>Soil 1</b>	<b>Soil 2</b>	<b>Soil 3</b>	<b>Soil 4</b>	<b>Biomass ash</b>
C (%)	1.07	3.75	3.10	3.39	11.65
N <sub>T</sub> (%)	0.09	0.34	0.25	0.31	0.21
pH <sub>w</sub>	4.80	5.05	4.63	4.74	11.31
Ca <sub>e</sub> (cmol <sub>(+)</sub> kg <sup>-1</sup> )	1.53	3.67	1.51	2.24	95.03
Mg <sub>e</sub> (cmol <sub>(+)</sub> kg <sup>-1</sup> )	0.41	0.71	0.52	0.64	3.26
Na <sub>e</sub> (cmol <sub>(+)</sub> kg <sup>-1</sup> )	0.25	0.12	0.21	0.35	12.17
K <sub>e</sub> (cmol <sub>(+)</sub> kg <sup>-1</sup> )	1.27	1.22	0.93	1.00	250.7
Al <sub>e</sub> (cmol <sub>(+)</sub> kg <sup>-1</sup> )	0.61	1.10	2.16	1.68	0.07
eCEC (cmol <sub>(+)</sub> kg <sup>-1</sup> )	4.08	6.82	5.33	5.92	361.2
Al <sub>o</sub> (mg kg <sup>-1</sup> )	855	4335	4950	5040	8323
Fe <sub>o</sub> (mg kg <sup>-1</sup> )	1150	2430	1390	2585	4233

485  
 486 pH<sub>w</sub> is pH measured in water; C is total organic carbon; N<sub>T</sub> is total nitrogen; eCEC is the effective cation exchange capacity (cmol<sub>e</sub>kg<sup>-1</sup>).  
 487 X<sub>e</sub> is exchangeable concentration of the element (cmol<sub>(+)</sub> kg<sup>-1</sup>); Al<sub>o</sub>, Fe<sub>o</sub>: extracted with ammonium oxalate (mg kg<sup>-1</sup>).  
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491 **Table 2.** Final concentrations of Fe and Al oxides in soils for each dose of biomass ash  
492 added to the soils studied. Average values (n=3) with coefficients of variation always  
493 <5%

	<b>Biomass ash dose</b>							
	<b>0 g kg<sup>-1</sup></b>		<b>6 g kg<sup>-1</sup></b>		<b>24 g kg<sup>-1</sup></b>		<b>48 g kg<sup>-1</sup></b>	
<b>Soil</b>	<b>Al<sub>o</sub></b>	<b>Fe<sub>o</sub></b>	<b>Al<sub>o</sub></b>	<b>Fe<sub>o</sub></b>	<b>Al<sub>o</sub></b>	<b>Fe<sub>o</sub></b>	<b>Al<sub>o</sub></b>	<b>Fe<sub>o</sub></b>
<b>Soil 1</b>	0.86	1.15	1.35	1.40	2.85	2.16	4.84	3.17
<b>Soil 2</b>	4.34	2.43	4.83	2.68	6.33	3.44	8.32	4.45
<b>Soil 3</b>	4.95	1.39	5.45	1.64	6.94	2.40	8.93	3.41
<b>Soil 4</b>	5.04	2.59	5.54	2.84	7.03	3.60	9.02	4.61

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Al<sub>o</sub>, Fe<sub>o</sub>: extracted with ammonium oxalate (mg kg<sup>-1</sup>)

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496 **Table 3.** Values of logarithm IC<sub>50</sub> (estimated values ± error) and R<sup>2</sup> from fits to logistic  
 497 model for the 3 antibiotics and 4 soils studied as a function of different biomass ash  
 498 (BA) doses added.

Antibiotic	BA dose	Soil 1		Soil 2		Soil 3		Soil 4	
		Log IC <sub>50</sub>	R <sup>2</sup>	Log IC <sub>50</sub>	R <sup>2</sup>	Log IC <sub>50</sub>	R <sup>2</sup>	Log IC <sub>50</sub>	R <sup>2</sup>
TC	0 g kg <sup>-1</sup>	2.26±0.09	0.975	2.71±0.06	0.985	2.51±0.05	0.989	2.68±0.08	0.973
	6 g kg <sup>-1</sup>	1.84±0.09	0.992	3.14±0.01	0.998	2.49±0.05	0.991	2.83±0.12	0.957
	24 g kg <sup>-1</sup>	-	-	2.80±0.1	0.963	3.35±0.03	0.992	2.85±0.18	0.941
	48 g kg <sup>-1</sup>	-	-	5.84±2.91	0.541	3.58±0.62	0.814	4.64±1.12	0.735
OTC	0 g kg <sup>-1</sup>	2.11±0.13	0.975	1.90±0.07	0.995	2.70±0.10	0.948	2.46±0.16	0.934
	6 g kg <sup>-1</sup>	2.77±0.10	0.971	2.96±0.21	0.906	1.80±0.05	0.997	2.49±0.07	0.983
	24 g kg <sup>-1</sup>	-	-	2.87±0.10	0.970	2.84±0.11	0.921	3.09±0.10	0.921
	48 g kg <sup>-1</sup>	-	-	4.75±1.53	0.695	3.64±0.13	0.929	3.35±0.46	0.818
CTC	0 g kg <sup>-1</sup>	1.82±0.12	0.985	2.00±0.08	0.982	2.15±0.08	0.965	2.42±0.03	0.996
	6 g kg <sup>-1</sup>	1.79±0.09	0.985	1.88±0.18	0.953	1.63±0.10	0.990	1.81±0.16	0.973
	24 g kg <sup>-1</sup>	-	-	2.17±0.12	0.987	3.08±0.07	0.953	3.99±0.26	0.943
	48 g kg <sup>-1</sup>	-	-	3.36±0.18	0.280	-	-	2.43±0.45	0.926

499 TC: tetracycline; OTC: oxytetracycline; CTC: chlortetracycline

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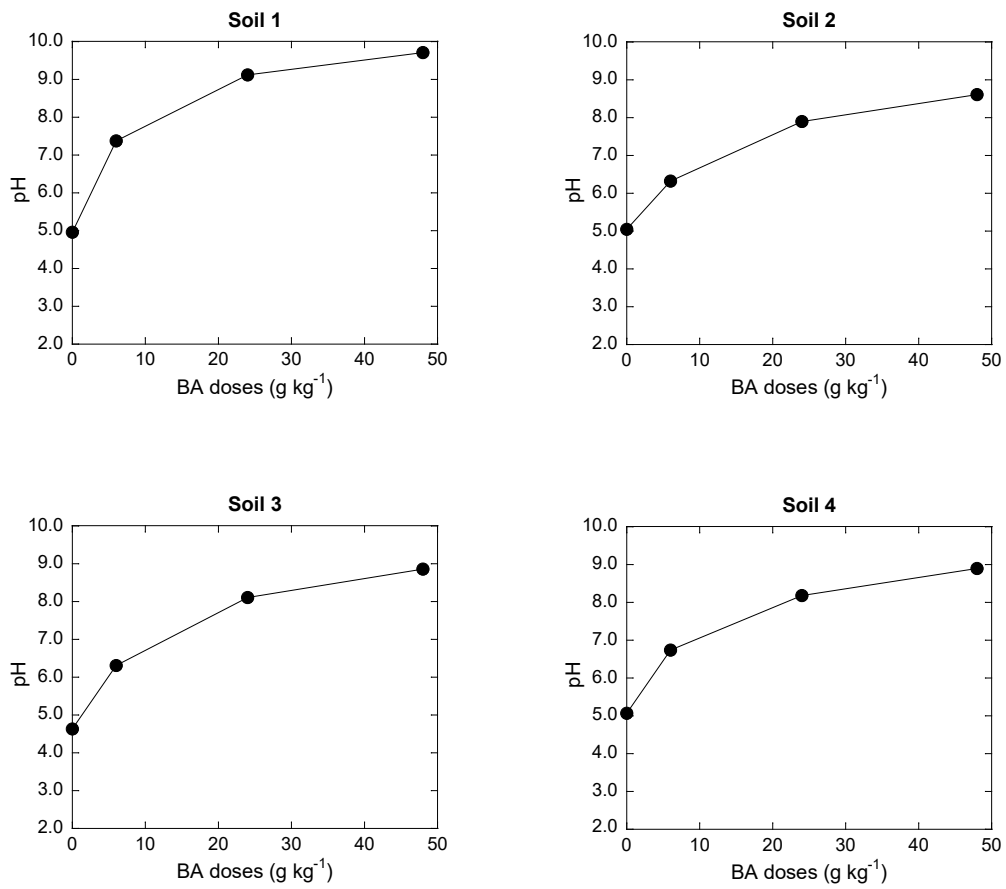
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**Table 4.** Percentage of each antibiotic species for different pHs measured in the amended soils. Average values (n=3) with coefficients of variation always <5%

Soil	BA dose (g kg <sup>-1</sup> )	pH	TC				OTC				CTC			
			TC <sup>+1</sup>	TC <sup>0</sup>	TC <sup>-1</sup>	TC <sup>-2</sup>	OTC <sup>+1</sup>	OTC <sup>0</sup>	OTC <sup>-1</sup>	OTC <sup>-2</sup>	CTC <sup>+1</sup>	CTC <sup>0</sup>	CTC <sup>-1</sup>	CTC <sup>-2</sup>
1	0	5.0	2.04	97.79	0.16	0.00	1.63	98.03	0.34	0.00	1.95	97.89	0.16	0.0
	6	7.4	0.01	70.44	29.36	0.19	0.00	52.74	45.93	1.32	0.01	71.39	28.42	0.18
	24	9.1	0.00	3.47	72.52	24.01	0.00	0.93	40.51	58.56	0.00	3.67	73.19	23.1
	48	9.7	0.00	0.52	42.91	56.57	0.00	0.09	14.79	85.12	0.00	0.55	44.02	55.4
2	0	5.1	1.63	98.17	0.21	0.00	1.30	98.28	0.43	0.00	1.56	98.25	0.20	0.0
	6	6.3	0.10	96.70	3.20	0.00	0.08	93.44	6.46	0.01	0.10	96.84	3.06	0.0
	24	7.9	0.00	42.63	56.20	1.17	0.00	24.97	68.76	6.27	0.00	43.78	55.12	1.10
	48	8.6	0.00	12.05	79.61	8.34	0.00	4.74	65.38	29.88	0.00	12.59	79.46	7.95
3	0	4.6	4.98	94.95	0.06	0.00	4.00	95.87	0.13	0.00	4.77	95.17	0.06	0.0
	6	6.3	0.10	96.70	3.20	0.00	0.08	93.44	6.46	0.01	0.10	96.84	3.06	0.0
	24	8.1	0.00	31.66	66.15	2.19	0.00	16.68	72.80	10.52	0.00	32.70	65.24	2.06
	48	8.9	0.00	5.90	77.83	16.26	0.00	1.86	51.33	46.81	0.00	6.21	78.19	15.6
4	0	5.1	1.63	98.17	0.21	0.00	1.30	98.28	0.43	0.00	1.56	98.25	0.20	0.0
	6	6.7	0.04	92.28	7.68	0.01	0.02	81.93	17.92	0.13	0.04	92.60	7.36	0.01
	24	8.2	0.00	26.74	70.33	2.93	0.00	13.34	73.32	13.34	0.00	27.69	69.54	2.77
	48	8.9	0.00	5.90	77.83	16.26	0.00	1.86	51.33	46.81	0.00	6.21	78.19	15.6

506 TC: tetracycline; OTC: oxytetracycline; CTC: chlortetracycline. The superscript (TC<sup>x</sup>.OTC<sup>x</sup> and CTC<sup>x</sup> is  
507 the valence of the antibiotic.  
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512 **Fig. 1.** pH variation as a function of Biomass Ash (BA) doses added to soils (0, 6, 12 and 48 g kg<sup>-1</sup>).

513 Average values (n=3) with coefficients of variation always <5%

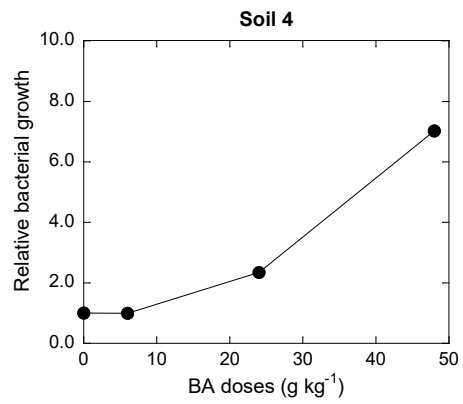
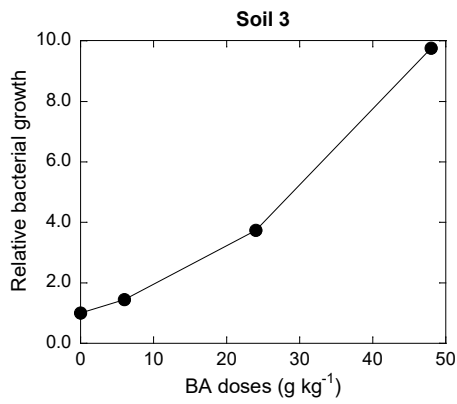
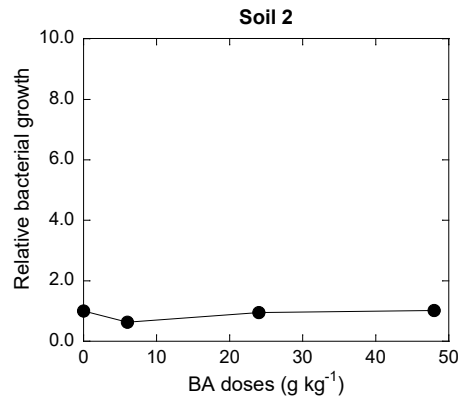
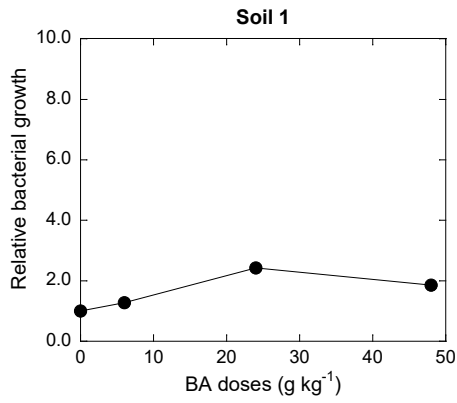
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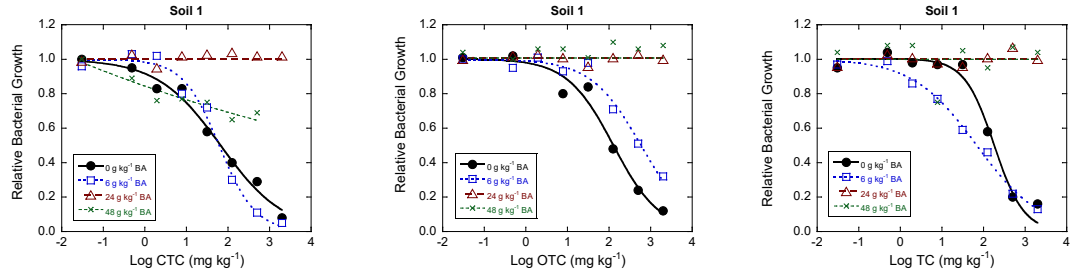


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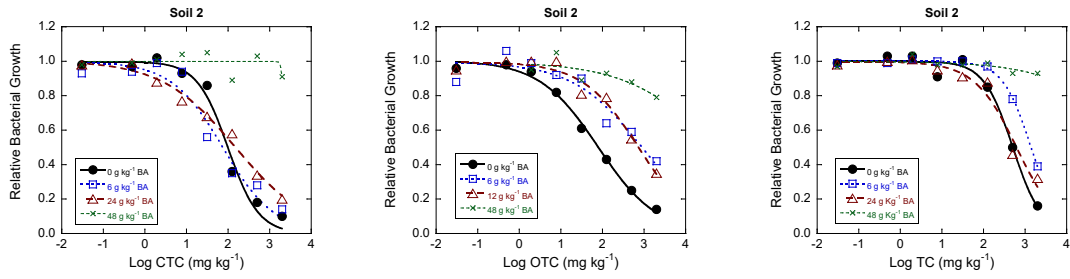
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521 **Fig. 2.** Relative bacterial growth in control samples (without antibiotics) as a function of different doses  
 522 of Biomass Ash added (0, 6, 24 and 48 g kg<sup>-1</sup>). Average values (n=3) with coefficients of variation always  
 523 <5%  
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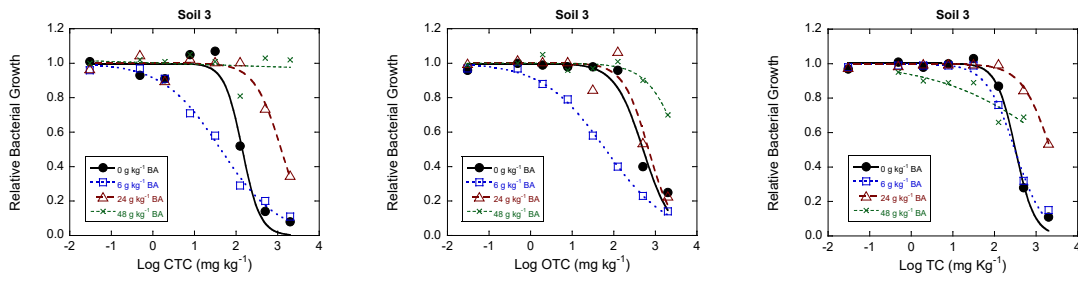
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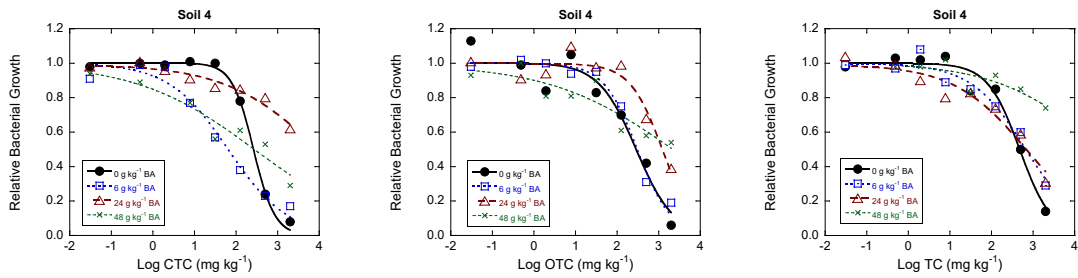
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**Fig. 3.** Bacterial community growth dose-response curves for 4 soils, 3 antibiotics [Tetracycline (TC),

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oxytetracycline (OTC) and chlortetracycline (CTC)], and 4 biomass ash (BA) doses. Average values

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(n=3) with coefficients of variation always <5%.