

SUPPLEMENTARY MATERIAL

Continuous UV-C/H₂O₂ and UV-C/Chlorine applied to municipal secondary effluent and nanofiltration retentate: removal of contaminants of emerging concern, ecotoxicity, and reuse potential

Fernando Rodrigues-Silva^a, Carla S. Santos^{b,c}, Joaquín A. Marrero^{c,d}, Rosa Montes^e, José Benito Quintana^e, Rosario Rodil^e, Olga C. Nunes^{c,d}, Maria Clara V.M. Starling^a, Camila C. Amorim^{a*}, Ana I. Gomes^{b,c*}, Vítor J.P. Vilar^{b,c*}

^a *Research Group on Environmental Applications of Advanced Oxidation Processes (GruPOA); Department of Sanitary and Environmental Engineering, The Federal University of Minas Gerais, Av. Antônio Carlos, 6627, 31270-901 Belo Horizonte, Minas Gerais, Brazil.*

^b *Laboratory of Separation and Reaction Engineering-Laboratory of Catalysis and Materials (LSRE-LCM), Faculty of Engineering University of Porto, Rua Dr. Roberto Frias, 4200-465, Porto, Portugal*

^c *Associate Laboratory in Chemical Engineering (ALiCE), Faculty of Engineering, University of Porto, Rua Dr. Roberto Frias, 4200-465, Porto, Portugal.*

^d *Laboratory for Process and Reaction Engineering, Environment, Biotechnology and Energy (LEPABE), Faculdade de Engenharia, Universidade do Porto, Rua Dr. Roberto Frias, 4200-465 Porto, Portugal*

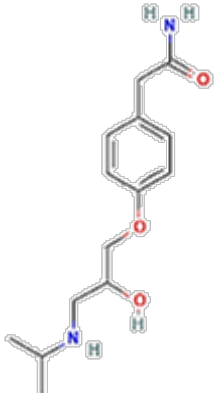
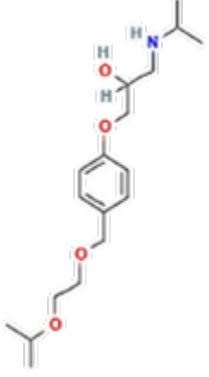
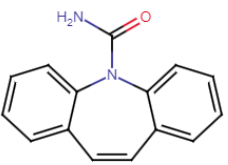
^e *Department of Analytical Chemistry, Nutrition and Food Sciences, Institute of Research on Chemical and Biological Analysis (IAQBUS), Universidade de Santiago de Compostela, Constantino Candeira S/N, 15782, Santiago de Compostela, Spain*

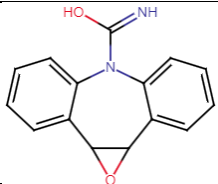
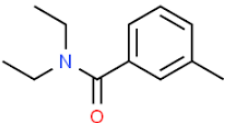
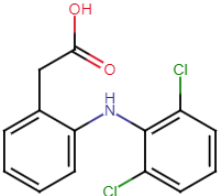
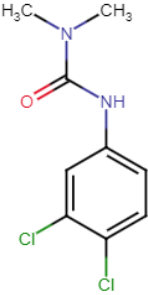
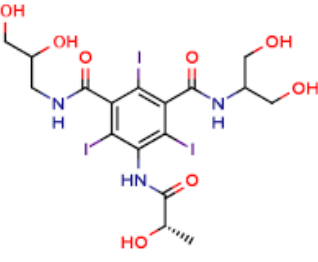
*Authors to whom all correspondence should be addressed (corresponding authors): camila@desa.ufmg.br (Camila C. Amorim); ana.isabelgomes@fe.up.pt (Ana I. Gomes); vilar@fe.up.pt (Vítor J.P. Vilar)

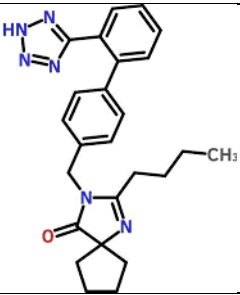
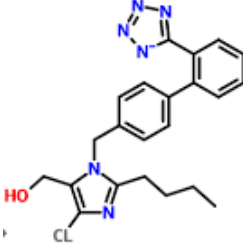
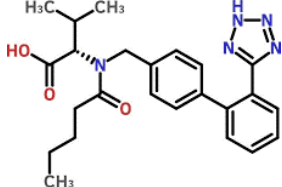
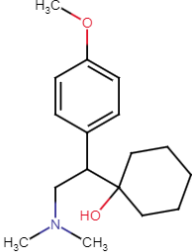
MATERIAL AND METHODS

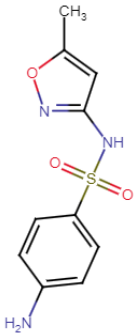
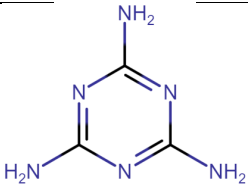
Chemicals

Table SM-1. List of the target CECs selected for testing and respective abbreviations and chemical composition.

CAS no.	Contaminant Acronym	Molecule structure	Organic functional group*	Chemical composition	MW (g mol ⁻¹)	pKa	Log K _{ow}	PNEC (µg L ⁻¹)
29122-68-7	Atenolol ATNL			C ₁₄ H ₂₂ N ₂ O ₃	266.336	9.6	0.43	150.0
66722-44-9	Bisoprolol BSPL		Aromatics: arene, benzene ring; CHN-containing: amine, amine secondary; CHO-containing: alkanol, ether;	C ₁₈ H ₃₁ NO ₄	325.443	9.5	2.20	92.0
298-46-4	Carbamazepine CBZ		Aromatics: arene, benzene ring; CH only: alkene; CHNO-containing: urea CHO-containing: carbonyl	C ₁₅ H ₁₂ N ₂ O	236.269	13.94	2.77	2.00

36507-30-9	Carbamazepine 10,11-epoxide CBZ-EPX		Aromatics: arene, benzene ring; CHN-containing: iminyl; CHO-containing: epoxide, ether	$C_{15}H_{12}N_2O_2$	252.268	-	1.97	2.57
134-62-3	N,N-diethyl-meta-toluamide DEET			$C_{12}H_{17}NO$	191.274	-	2.50	88.0
15307-86-5	Diclofenac DCF		Aromatics: aniline, arene, benzene ring; CHN-containing: amine, amine secondary; CHO-containing: carbonyl, carboxylic acid; Halogen-containing: aryl chloride, aryl halide, leaving group	$C_{14}H_{11}Cl_2NO_2$	296.148	4.15	4.26	0.05
330-54-1	Diuron DRN		Aromatics: arene, benzene ring; CHNO-containing: urea; CHO-containing: carbonyl; Halogen-containing: aryl chloride, aryl halide, leaving group	$C_9H_{10}Cl_2N_2O$	233.1	13.55	2.53	0.07
73334-07-3	Iopromide IOP		Aromatics: arene, benzene ring; CHNO-containing: carbomaxamide, carboxamide secondary, carboxamide tertiary CHO-containing: alkanol, carbonyl, 1,2-diol, ether; Halogen-containing: aryl halide, aryl iodide, leaving	$C_{18}H_{24}I_3N_3O_8$	791.112	9.9	-0.44	0.14

			group						
138402-11-6	Irbesartan ISTN		Aromatics: arene, azarene, benzene ring, heteroarene CHN-containing: amidine, amine, aminyl; CHNO-containing: carboxamide, carboximide tertiary; CHO-containing: carbonyl;	$C_{25}H_{28}N_6O$	428.53	4.12	5.39	700	
124750-99-8	Losartan LSTN		Aromatics: arene, azarene, benzene ring, heteroarene; CHN-containing: amine, amine tertiary; CHO-containing: alkanol; Halogen-containing: aryl chloride, aryl halide, leaving group	$C_{22}H_{23}ClN_6O$	422.91	4.9	5.00	0.0019	
137862-53-4	Valsartan VSTN			$C_{24}H_{29}N_5O_3$	435.519	4.73	5.27	560.0	
93413-69-5	Venlafaxine VLX		Aromatics: arene, benzene ring; CH only: alkene; CHN-containing: amine, amine tertiary; CHO-containing: alkanol, ether;	$C_{17}H_{27}NO_2$	277.402	9.4	2.74	0.88	

723-46-6	Sulfamethoxazole SMX		Aromatics: aniline, arene, azarene, arene, benzene ring, heteroarene; CHN-containing: amine, amine primary; S-containing: sulfonamide;	$C_{10}H_{11}N_3O_3S$	253.279	3.92	0.79	0.60
108-78-1	Melamine MLN		Aromatics: aniline, arene, azaarene, heteroarene CHN-containing: amine, amine primary;	$C_3H_6N_6$	126.12	5.00	-0.59	3.54

*Organic functional group was determined by an open online tool – ACE functional group finder [<https://epoch.uky.edu/ace/public/fnalGroups.jsp>].

CECs Quantification by Liquid Chromatography-Tandem Mass Spectrometry (LC-MS/MS)

Analysis was performed by direct injection of 45 μL of the sample into the chromatographic system Acquity UPLC® H class interfaced to a Xevo TQD triple quadrupole mass spectrometer (Waters, Milford, MA, USA). Separation took place in a Luna C18 100A column (50 mm \times 2 mm, 3 μm particle size) supplied by Phenomenex (Torrance, CA, USA), with constant temperature of 30 $^{\circ}\text{C}$ and 0.2 mL min^{-1} flow rate. The mobile phases consisted of (A) 1 mM ammonium fluoride in ultrapure water and (B) 1 mM ammonium fluoride in MeOH. The gradient was as follows: 0-1 min, 0% B; 1-8 min, linear gradient to 100% B; 7-13 min, 100% B and finally 13-20 min, 0% B. Nitrogen was used as nebulising and drying gas and Argon was used as collision gas. The analyses were determined in the ESI positive and negative polarities, and the acquisition mode was multiple-reaction monitoring (MRM). Two MRM transitions were used as a quantifier (Q) and qualifier (q) for each compound (Table SM-2). The quantification of CECs was performed by the standard addition method spiking each real matrix with the target CECs at 5 different concentration levels (LOQ-50 $\mu\text{g/L}$). Linearity, in terms of R^2 was higher than 0.990 for all analytes in both matrices and repeatability, in terms of RSD (%) was lower than 15 % (5 consecutive injections of the 10 $\mu\text{g/L}$ spiked level in each matrix). The method limits of quantification (LQ) are shown in Table SM-3 and were estimated by extrapolating the signal to noise (S/N) obtained from the chromatographic peaks at the specific native concentration in real samples (or the lowest addition level if the CECs were not present) to a S/N of 10.

Table SM-2. Instrumental LC-MS/MS parameters.

CECs	Retention time (min)	Precursor ion (<i>m/z</i>)	Product ion (<i>m/z</i>)	Collision energy (V)	Cone voltage (V)	ESI mode
ATNL	5.7	267	145 (Q)	30	38	+
			190 (q)	16		
BSPL	7.5	326	116 (Q)	20	45	+
			204 (q)	18		
CBZ	7.9	237	194 (Q)	20	40	+
			179 (q)	38		
CBZ-EPX	7.1	253	180 (Q)	25	39	+
			210 (q)	12		
DCF	8.7	294	250 (Q)	10	33	-
		296	252 (q)	10		
DEET	8.2	192	119(Q)	15	25	+
			91(q)	25		
DRN	8.4	233	72 (Q)	20	30	+
			160 (q)	20		
IOP	5.2	792	559 (Q)	25	50	+
			573 (q)	15		
ISTN	8.6	434	179 (Q)	65	50	-
			350 (q)	28		
LSTN	8.3	421	127 (Q)	30	30	-
			157 (q)	25		
VSTN	8.4	434	179 (Q)	25	35	+
			350 (q)	20		
MLN	1.5	127	85 (Q)	15	45	+
			69 (q)	20		
SMX	6.2	254	92 (Q)	25	35	+
			156 (q)	15		
VLX	7.9	278	58 (Q)	20	25	+
			260 (q)	10		

Note: Capillary voltages: 3.5 and 1.5 kV in positive and negative polarities, respectively. Source temperature: 350 °C. Desolvation gas (N₂) flow: 650 L h⁻¹. Cone gas (N₂) flow: 5 L h⁻¹. Collision energy (CE) and cone voltage (CV) were adjusted for every transition. (Q) quantifier transition; (q) qualifier transition.

Table SM-3. LC-MS/MS method LQs in municipal secondary effluent and nanofiltration retentate

CECs	Municipal Secondary Effluent	Nanofiltration Retentate
	LQ ($\mu\text{g L}^{-1}$)	LQ ($\mu\text{g L}^{-1}$)
ATNL	0.15	0.18
BSPL	0.15	0.22
CBZ	0.03	0.05
CBZ-EPX	0.02	0.10
DCF	0.25	0.40
DEET	0.03	0.05
DRN	0.08	0.08
IOP	0.80	1.0
ISTN	0.20	0.30
LSTN	0.80	1.0
VSTN	0.44	0.17
MLN	0.75	1.0
SMX	0.10	0.10
VLX	0.15	0.20

RESULTS AND DISCUSSION

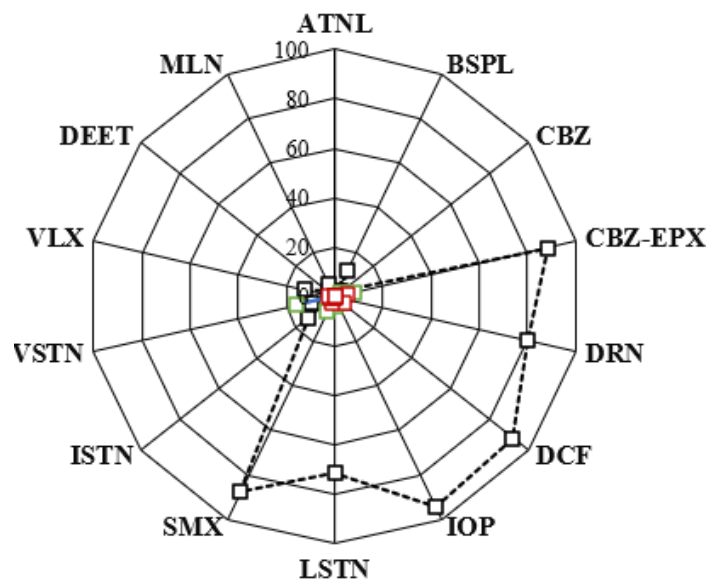


Figure SM-1. Removal percentage for the 14 target CECs spiked in MSE ($10 \mu\text{g L}^{-1}$ for each) under UV-C irradiation (3.3 kJ L^{-1}) without oxidant dosage ($-\square-$) and under dark conditions with H_2O_2 ($-\square-$) or NaClO ($-\square-$) dosage.

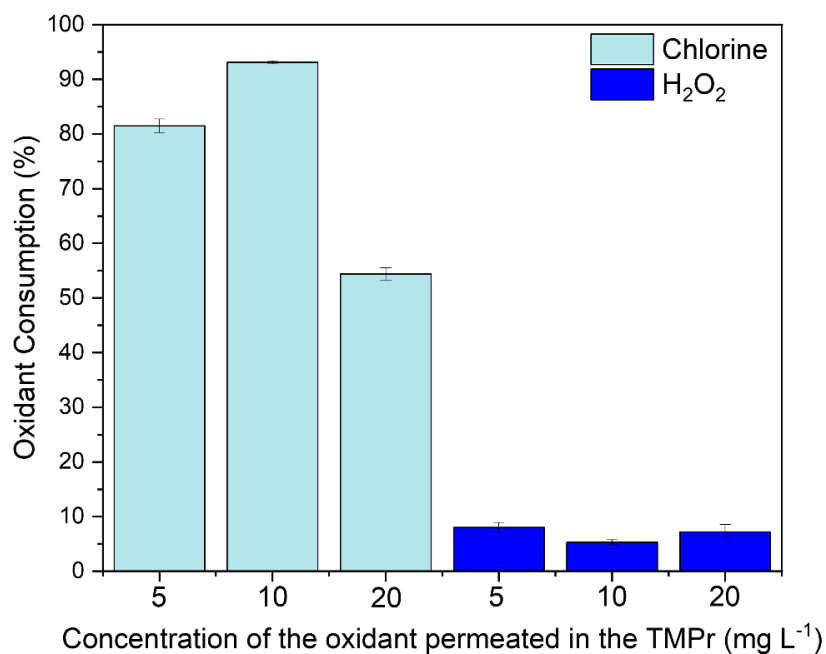


Figure SM-2. Consumption of the oxidants for the photo-treatment tests performed with MSE. Conditions: treatment time 3.4 min, UV-C dose of 3.3 kJ L^{-1} .

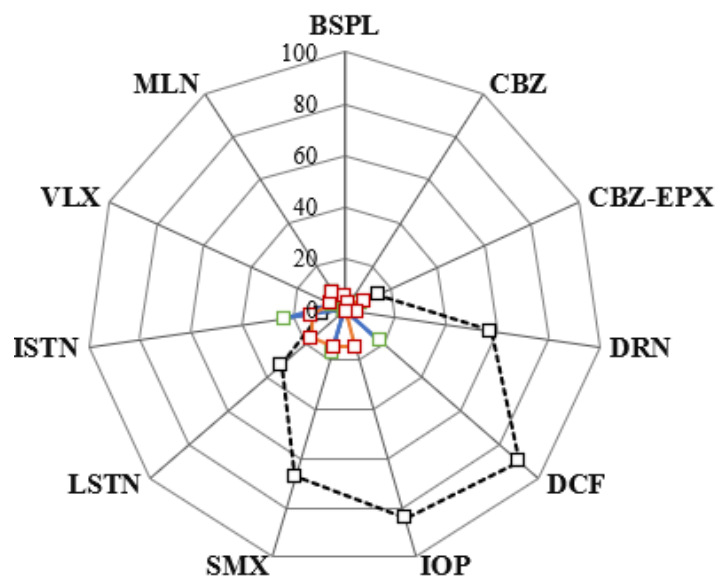


Figure SM-3. Removal percentage for the 11 target CECs present in NF_R under UV-C irradiation (3.3 kJ L⁻¹) without oxidant dosage (—□—) and under dark conditions with H₂O₂ (—□—) or NaClO (—□—) dosage.

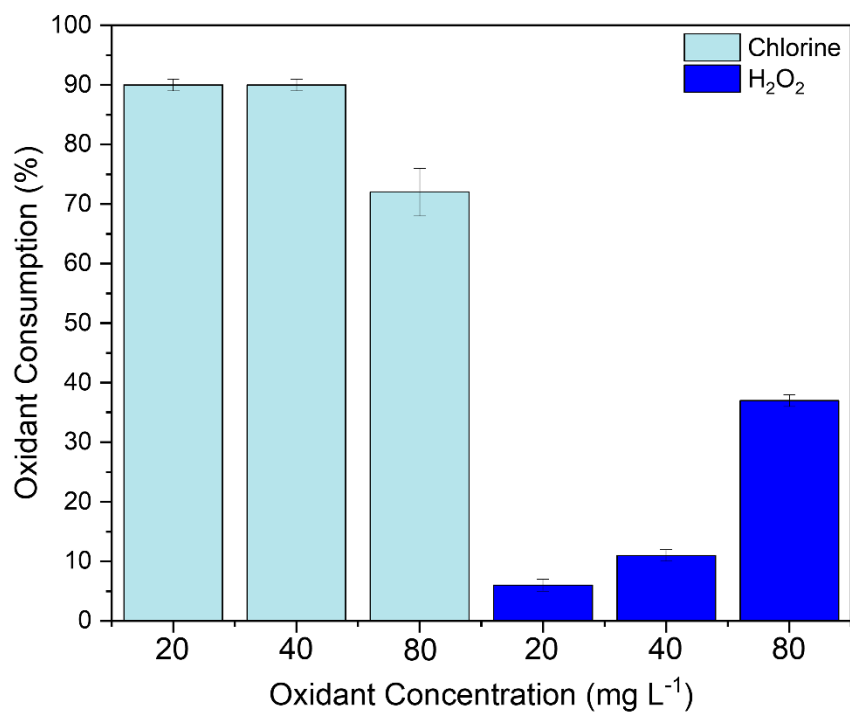


Figure SM-4. Consumption of the oxidants for the photo-treatment tests performed with NF_R. Conditions: treatment time 3.4 min, UV-C dose of 3.3 kJ L⁻¹.

Table SM-4. Physicochemical characterization of MSE before and after standalone UV-C (test #1); standalone H₂O₂ at 20 mg L⁻¹ (test #2) and Chlorine at 20 mg L⁻¹ (test #3); UV-C/H₂O₂ with H₂O₂ permeation at 5 mg L⁻¹ (test #4), 10 mg L⁻¹ (test #5), and 20 mg L⁻¹ (test #6); UV-C/Chlorine with free chlorine permeation at 5 mg L⁻¹ (test #7), 10 mg L⁻¹ (test #8), and 20 mg L⁻¹ (test #9).

Sample	COD	DIC	DOC	Abs. ₂₅₄	SUVA ₂₅₄	Turb.	pH	Conduc.	TSS	Cl ⁻	NO ₂ ⁻	NO ₃ ⁻	SO ₄ ²⁻	PO ₄ ²⁻	Na ²⁺	NH ₄ ⁺	K ⁺	Mg ²⁺	Ca ²⁺
Unit	mg L ⁻¹			-	Lmg ⁻¹ m ⁻¹ ₁	NTU	-	μS cm ⁻¹	mg L ⁻¹										
MSE	74	39	23	0.231	0.984	3.1	7.70	846	14	130	2	13	43	5	112	14	18	8	31
Test #1	79	38	24	0.208	0.870	3.1	7.68	864	7	136	3	15	44	6	117	15	18	9	34
Test #2	68	38	23	0.226	0.995	3.7	7.65	864	10	128	3	12	42	6	113	15	17	9	33
Test #3	74	36	24	0.243	0.999	3.1	7.98	910	12	151	3	19	44	6	127	15	18	9	34
Test #4	77	36	24	0.170	0.708	1.1	7.85	857	4	132	3	19	46	6	114	13	18	9	34
Test #5	76	36	23	0.164	0.701	1.1	7.61	867	5	133	3	19	43	6	112	13	18	9	34
Test #6	70	35	23	0.171	0.735	1.4	7.54	837	6	132	3	21	42	6	113	12	18	9	33
Test #7	67	37	24	0.196	0.829	1.4	7.87	879	2	142	3	15	43	5	118	14	17	9	33
Test #8	69	36	24	0.190	0.793	1.8	7.93	936	3	151	3	16	43	5	125	14	18	9	33
Test #9	72	36	24	0.191	0.810	1.2	8.02	920	4	154	3	18	46	6	129	13	18	9	34

Note: MSE – Municipal secondary effluent; COD = chemical oxygen demand; DIC = dissolved inorganic carbon; DOC = dissolved organic carbon; Turb. = turbidity; Conduc. = electrical conductivity; TSS = total suspended solids.

Table SM-5. Physicochemical characterization of NF_R before and after standalone UV-C (test #10); standalone H₂O₂ at 80 mg L⁻¹ (test #11) and Chlorine at 80 mg L⁻¹ (test #12); UV-C/H₂O₂ with H₂O₂ permeation at 20 mg L⁻¹ (test #13), 40 mg L⁻¹ (test #14), and 80 mg L⁻¹ (test #15); UV-C/Chlorine with free chlorine permeation at 20 mg L⁻¹ (test #16), 40 mg L⁻¹ (test #17), and 80 mg L⁻¹ (test #18).

Sample	COD	DIC	DOC	Abs. ₂₅₄	SUVA ₂₅₄	Turb.	pH	Conduc.	TSS	Cl ⁻	NO ₂ ⁻	NO ₃ ⁻	SO ₄ ²⁻	PO ₄ ²⁻	Na ²⁺	NH ₄ ⁺	K ⁺	Mg ²⁺	Ca ²⁺
Unit	mg L ⁻¹			-	Lmg ⁻¹ m ⁻¹	NTU	-	μS cm ⁻¹	mg L ⁻¹										
NF _R	206	65	51	1.168	2.28	1.7	8.08	2204	9	561	27	135	636	79	751	57	111	44	224
Test #10	164	58	53	1.133	2.13	2.7	8.06	2092	5	625	31	165	639	86	543	52	115	46	234
Test #11	151	61	53	1.179	2.22	3.4	8.04	2031	7	478	32	172	627	85	541	56	117	46	237
Test #12	144	57	53	1.134	2.16	3.0	8.11	2297	13	481	38	152	630	87	651	47	114	47	239
Test #13	161	61	54	1.063	1.98	2.9	8.07	2117	30	480	38	155	641	84	541	55	117	46	234
Test #14	122	60	52	1.034	2.00	2.5	7.92	2106	11	482	38	161	639	85	543	55	115	46	233
Test #15	151	60	51	1.028	2.00	3.1	7.87	2090	8	481	37	157	637	86	544	54	115	47	237
Test #16	165	67	52	1.104	2.12	2.6	7.98	2002	8	549	32	136	649	82	793	58	110	43	225
Test #17	135	66	51	1.103	2.22	3.2	8.21	2135	16	611	33	140	633	82	597	61	113	45	232
Test #18	140	67	51	1.058	2.07	3.4	8.26	2290	18	483	40	157	629	82	637	59	117	46	233

Note: NF_R = nanofiltration retentate from urban wastewater; COD = chemical oxygen demand; DIC = dissolved inorganic carbon; DOC = dissolved organic carbon; Turb. = turbidity; Conduc. = electric conductivity; TSS = total suspended solids.