



Influence of magnetic nanoparticles on the mechano-magnetic response of wet-spun sodiumalginate-nanocellulose filaments

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Abstract Hybrid filaments are of growing interest for a wide range of applications, including those that require stimuli-responsiveness. In this study we developed magnetic filaments by combining the properties of inorganic nanoparticles with the low density, flexibility and morphological features of 2,2,6,6-tetramethylpiperidine-1-oxyl (TEMPO)-oxidized cellulose nanofibrils (TOCNF). The hybrid filaments were synthesized by wet spinning of TOCNF using sodium alginate (SA) adjuvant in a hydrogel containing magnetite (Fe_3O_4) nanoparticles (NPs) formed *in-situ* by nucleation and grow. The relationship between synthesis conditions and filament mechanical and magnetic properties were investigated at NP loading as high as 25%. Saturation magnetization of 1.60,

11.31, 19.41, and 33.25 emu/g Fe_3O_4 were measured at 5, 10, 15, and 25% NPs with a penalty in filament tensile strength which nevertheless reached at least 118 GPa along with low magnetite crystal orientation. Such high strength is rarely reported and found to depend on cellulose crystal orientation. The magnetic filaments were found suitable to replace traditional magnetic systems but add to the opportunity to develop flexible microwave adsorption textiles, artificial muscles, and micro-sensors.

Keywords Filaments · Nanocellulose · Wet spinning · Coagulation · Magnetic filaments · Coaxial filaments

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Introduction

Depending on the nature and size of magnetic nanoparticles (MNPs), they show superparamagnetic (SPM), ferromagnetic (Ferro), ferrimagnetic (Ferri), or antiferromagnetic (anti-Ferro) behavior, which are of interest in applications, such as magnetic resonance imaging (MRI), targeted drug delivery and in biological separation (Liao and Chen 2002; Chen et al. 2013; Sahoo et al. 2013). In combination with other substances, MNPs produce versatile multi-functional materials. Of particular interest are magnetite (Fe_3O_4) nanoparticles (NPs) which can be customized in size, shape, and function by incorporating ad hoc designed surface properties (Li et al. 2011; Liu et al. 2013). Moreover, they can be scaled up for industrial production via cost-effective and eco-friendly processes such as coprecipitation (Blaney 2007). Owing to their thermal stability (Li et al. 2011), Fe_3O_4 NPs have a relevant role in diverse applications, including metals separation, data storage media, microwave adsorption, and shape memory (Liu et al. 2005; Schmidt 2006; Yavuz et al. 2006; Mincheva et al. 2008; Li et al. 2011).

Biopolymers such as cellulose, lignin, chitosan or alginate (Verma and Fortunati 2019) can be sourced from residues from the forest, food or marine industries, and represent an alternative to the petroleum-based products. Cellulose, in particular, is found in plant structures composed of ordered and disordered domains and is isolated in the form of cellulose nanocrystals (CNCs) and nanofibrils (CNFs) (Moon et al. 2011). The latter can be produced by chemical, physical, or physicochemical pretreatment methods. Treatment with 2,2,6,6-Tetramethylpiperidine-1-oxyl radical (TEMPO) produces anionic CNFs (referred to TOCNFs) with very small width (3–4 nm) and, at least a few microns in length. TOCNF form lightweight, strong structures that benefit from the presence of carboxylate groups ($-\text{COO}$), which impart electrostatic stabilization (Saito et al. 2009; Isogai et al. 2011; Levanič et al. 2020).

Alginate, extracted from brown seaweeds is composed of β -D-mannuronate (M) and α -L-guluronate (G) units linked in given block sequences (M-M, G-G, and M-G) and is widely used in the pharmaceutical, cosmetic, and food industries. In addition, alginates are considered in environmental remediation due to the effect of gelation, biocompatibility,

non-toxicity, adsorption capability and low cost (He et al. 2020).

Hybrid biobased materials, for instance, based on TOCNF and alginates, can achieve unique properties and potential advantages arising from the inorganic phase. The latter contribute to enhanced thermal stability, refractive index, hardness, optical or stimuli responsiveness (Chiellini and Solaro 2003; Khan et al. 2023). Alginates are being used as scaffolding materials to incorporate inorganic NPs or biologically active substance for medical applications due to their porosity and high surface-to-volume ratio (Bhattarai et al. 2006). Likewise, nanocellulose is used as natural bio-template combined with inorganic nanoparticles in the shape of membranes (Zheng et al. 2013), nanopaper (Li et al. 2013), filaments (Wang et al. 2019, 2020), and others (Shi et al. 2013). The resulting materials are known for their excellent conductivity or magnetic properties combined with outstanding mechanical performance and flexibility. The combination of these characteristics can introduce novel stimuli-responsive properties into flexible electrodes, displays, drug release devices or even in implantable biosensors (Shi et al. 2013).

Magnetic filaments, combining the magnetic response of MNPs and the lightness, transparency, and flexibility from biopolymers such as alginate and nanocellulose. Efforts have been made to enhance filaments homogeneity by embedding small ($D=30$ nm) SPM NPs in polymeric nanofibers (Wang et al. 2008; Miyauchi et al. 2011; Zhang et al. 2017) via homogenous spinning and extrusion. MNPs and polymeric dispersions have been processed by coaxial spinning (Mincheva et al. 2008), where the polymer solution was placed in the shell and MNPs at the core. Wet spinning, on the other hand, is recognized to produce filaments at industrial scale (Zhou et al. 2017; Kim et al. 2019) and is suitable to produce filaments with a uniaxial orientation that ensures enhanced mechanical performance (Iwamoto et al. 2011; He et al. 2020).

Recently, alginate filaments produced via wet spinning were produced for several applications (Fan et al. 2005, 2007; Sa and Kornev 2011; Zhao et al. 2019). The sodium alginate spinning dope was extruded into a coagulation bath containing CaCl_2 to form a filament with sufficient mechanical strength. Alginate has also been blended with other polymers and NPs to improve their properties in

the form of filaments. Recently, alginate filaments were reinforced with cellulose nanofibrils proving a great potential in the biomedical field (Park et al. 2021). The addition of functional NPs, such as gold nanoclusters embedded into alginate filaments, imparts not only high mechanical strength but also fluorescence, opening up possibilities for technological applications as anticounterfeiting label in the textile field (He et al. 2020).

Despite the progress made in the area of filament spinning, the incorporation of SPM NPs with sodium alginate and cellulose nanofibrils is still unexplored. In this work, we introduce the synthesis of Fe_3O_4 NPs by co-precipitation, using both *in-situ* (synthesized in a TOCNF matrix) and *ex-situ* (without TOCNF matrix) methods to prepare magnetic filaments. The effect of MNPs synthesized *in-situ* on the nanofibrils level was accessed. We further investigated the impact of the NP synthesis method and concentrations on the mechanical, magnetic, and morphological properties of the developed magnetic filaments (Mono and Coaxial).

Materials and methods

Materials

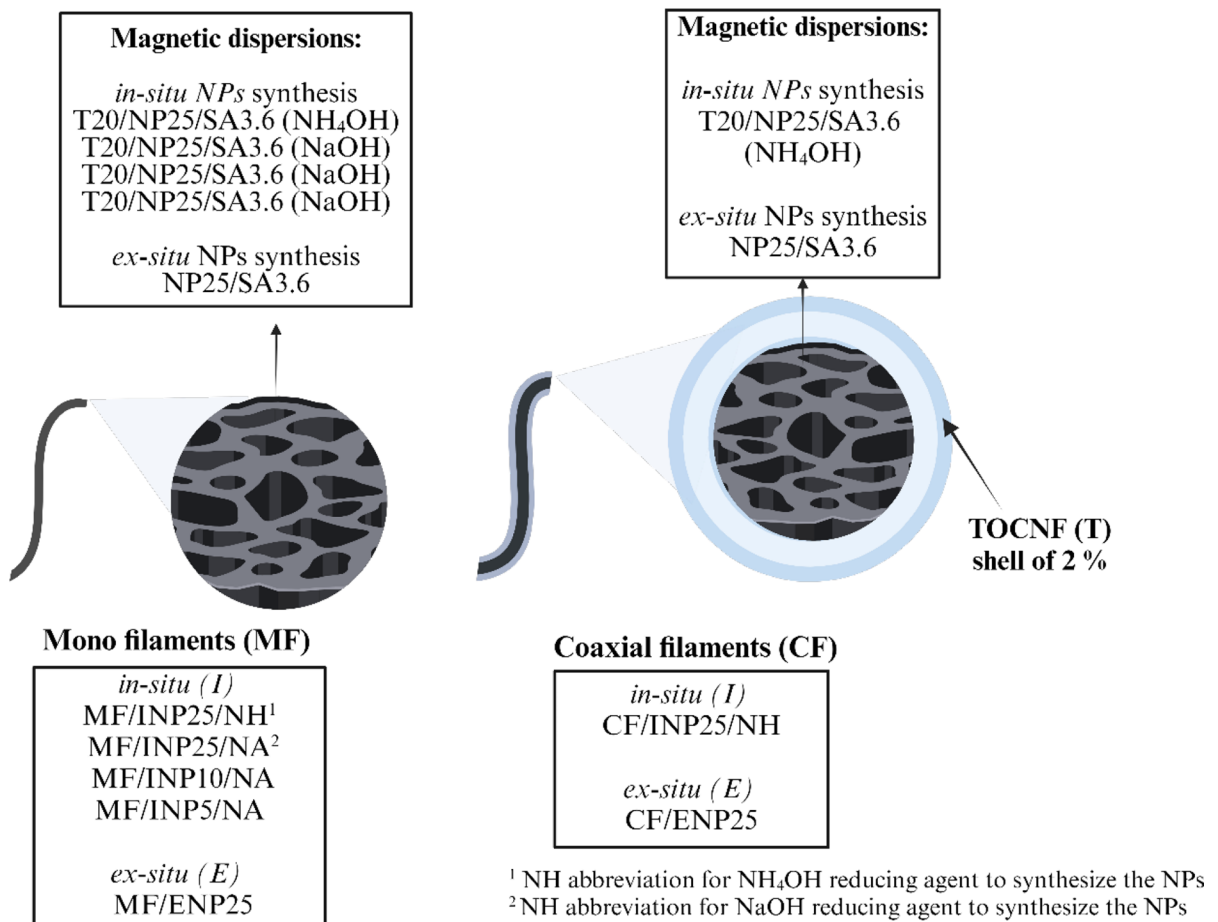
All chemicals were used without further purification. Iron (III) chloride hexahydrate ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$, 99%) was obtained from Alfa Aesar (Madrid, Spain), iron (II) sulphate heptahydrate ($\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$, 99%), calcium chloride (CaCl_2), ammonium hydroxide (NH_4OH , 25 wt%), sodium hydroxide (NaOH), absolute ethanol, and sodium alginate were purchased from Sigma (Saint Louis, MO, USA). Never dried bleached hardwood (birch) pulp was oxidized by 2,2,6,6-tetramethylpiperidine-1-oxyl (TEMPO) at pH 10, then washed with deionized water. The TEMPO-oxidized fibers were further microfluidized through one pass high-pressure microfluidization (Microfluidics Corp., USA), yielding TEMPO-oxidized nanofibrils (TOCNF). The obtained TOCNF (1.6 wt% of dry content) had a carboxylic group content of 0.6 mmol g^{-1} . TOCNF aqueous suspensions were prepared at 1.23% and 2% solid content.

Fe_3O_4 NPs prepared *in-situ* and *ex-situ*

Fe_3O_4 NPs were synthesized through two methods: *in-situ* synthesis within a suspension of TOCNF, and *ex-situ* (without TOCNF) synthesis via the coprecipitation method. In the *in-situ* method, 5.4 g of $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ and 3.3 g of $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ were added to a 250 mL aqueous suspension containing 40 wt% TOCNF based on the total dry mass. After complete dissolution of the iron salts, the mixture was heated at a constant temperature of 80°C for approximately 30 min under mechanical stirring and N_2 flow. Thereafter, 15 mL of NH_4OH (25 wt%) was added dropwise, and the solution was continuously stirred for an additional 30 min. The reaction was stopped, allowing the solution to cool to room temperature, followed by thorough rinsing with water and ethanol (using a permanent magnet). The percentage of NPs and TOCNF in dispersion ($V = 150 \text{ mL}$) was determined by thermogravimetric analysis (TGA), revealing that 25% of the material comprised NPs, while 20% consisted of TOCNF in solution.

In another method, 15 mL of NaOH (5 M) was used, and different weight ratios of iron salts ranging from 5.4 g – 0.24 g for $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ and from 3.3 g to 0.15 g for $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ were added to obtain various concentrations of NPs in solution (accounting for the loss of NPs during cleaning). This method followed the same procedure as described above, resulting in NP concentrations of 25, 10, and 5%, with 20% TOCNF in each of the magnetic dispersions ($V = 150 \text{ mL}$). In all the magnetic *in-situ* dispersions it was added 5.4 g sodium alginate (SA) and were left agitating for 4 h at room temperature. The final magnetic *in-situ* dispersions containing TOCNF, SA, and MNPs, were stored at 4°C until their use and were referred as: T20/NP25/SA3.6 (NH_4OH), T20/NP10/SA3.6 (NaOH), T20/NP5/SA3.6 (NaOH), and T20/NP25/SA3.6 (NaOH). The T, NP, and SA are used as abbreviations for the TOCNF (T), nanoparticle (NP), and sodium alginate (SA), respectively, and the corresponding % loading after the latter component is indicated by its percentage in suspension T (20%), NP (25, 10, and 5%) and SA (3.6%). After the label is stated, the reducing agent used in the preparation of the magnetic NPs.

The *ex-situ* preparation of Fe_3O_4 NPs, is basically done following the same procedure as the *in-situ* synthesis using NH_4OH as the reducing agent



Scheme 1. Magnetic dispersions prepared per type of filaments, Mono (MF) and Coaxial (CF)

in the absence of TOCNF, but with minor differences. The iron salts (5.4 g of FeCl₃·6H₂O and 3.3 g of FeSO₄·7H₂O) were added to 80 mL MilliQ. After their complete dissolution, the solution was heated at a constant temperature of 80 °C for approximately 30 min under mechanical stirring and N₂ flow. The reaction was stopped, let to cool down at room temperature, before being cleaned with MilliQ water (using a permanent magnet). The total magnetic content was determined by TGA, and a magnetic dispersion with 25% NPs was prepared in 150 mL MilliQ water. After which it was added 5.4 g of SA and left agitating for 4 h at room temperature. The magnetic *ex-situ* dispersion was also stored at 4 °C until use, and referred as: NP25/SA3.6 (comprised of 25% NPs and 3.6% SA).

Filament wet spinning

The prepared magnetic dispersions were degassed (3 min, 2500 rpm) via a planetary centrifugal mixer (THINKY ARE-250). The homogenous dispersions were then loaded into a 20 mL plastic syringe attached to a plastic tube (44.5 cm length and 6 mm inner diameter) that ended in a coaxial needle (inner diameter of 1.2 mm, outer diameter of 2.1 mm, and a length of 3.7 cm).

Two types of filaments, mono and coaxial (Scheme 1), were fabricated by immersing the coaxial needle in a coagulation bath (1 M CaCl₂) at a fixed spinning rate, 10 mL/min. The filaments remained in the coagulation bath for 20 min to allow time for cross-linking. Subsequently, the filaments underwent

two washing steps with distilled water (first step for 10 min and second step for 20 min) and collected and air dried under fixed ends to prevent shrinkage.

The drying time varied based on the filament type: coaxial filaments with a TOCNF shell (2% dry weight) took approximately 2.5 h to dry, while the mono filaments required approximately 1 h. Mono filaments were prepared from all magnetic dispersions (*in-situ* and *ex-situ*), while only coaxial filaments were prepared using the *in-situ* and *ex-situ* magnetic dispersions of T20/NP25/SA3.6 (NH₄OH) and NP25/SA3.6, respectively. Additionally, besides the magnetic filaments, reference mono and coaxial filaments (without magnetic NPs) were spun.

The magnetic wet-spun mono (M) and coaxial (C) filaments (F) containing the given NP amounts, their *in-situ* (I) and *ex-situ* (E) synthesis method, were therein referred to as MF/INP25/NH, MF/INP25/NA, MF/INP10/NA, MF/INP5/NA, MF/ENP25, CF/INP25/NH, and CF/ENP25 (where the abbreviations refer to MF (mono filament), CF (coaxial filament), INP (*in-situ* nanoparticles), ENP (*ex-situ* nanoparticles), NH (refers to NH₄OH reducing agent), and NA (refers to NaOH reducing agent), respectively).

Nanoparticles and filaments characterization

Filament morphology

The morphology and NP size were assessed by transmission electron microscopy (TEM) by using a JEOL JEM-1011 microscope (JEOL, Tokyo, Japan). Prior to the measurements, a diluted dispersion of each nanoparticle was prepared. Approximately 10 μ l of the diluted particle dispersion was deposited onto a copper grid with a thin layer of carbon film (Formvar films) and was left to dry under a white light bulb.

The filaments morphology and cross sections at break were observed by scanning electron microscopy (SEM) using a FESEM ZEISS Ultra plus microscope (Germany) operated at 20 kV with a working distance of 8.5 mm. Before imaging, pieces of the filaments were cut, by bending the frozen filaments in liquid N₂. The filaments were placed horizontally and vertically (cross section images) in 3 nm-Platinum/palladium (Pt/Pd) sputter coater targets. For detection of elements (Fe and C), energy-dispersive X-ray spectroscopy (EDS) spectra were recorded using the

high-resolution microscope with the EDS detector (model INCA-X act, Oxford).

Chemical and structural characterization

TOCNF, SA, magnetic dispersions and filaments was evaluated by Fourier transform infrared spectra (FTIR) and recorded on a Varian FT-IR 670 (Varian, Palo Alto, CA, USA) spectrophotometer in of range 400–4000 cm⁻¹. The structural properties were evaluated by X-ray diffraction (XRD) using a Philips PW1710 diffractometer (Panalytical, Callo End, UK), a “PW1820/00” vertical goniometer, and an “Enraf Nonius FR590” generator at 40 kV and 30 mA. The X-rays were obtained from a sealed Cu tube, and the radiation was monochromatized with a graphite monochromator ($\lambda (K_{\alpha 1}) = 1.5406 \text{ \AA}$). Measurements were collected in the angular range of 10–80° with a step size of 0.02° and a step time of 2 s per step. The mean particle size, dispersity (\bar{D}) and zeta potential of the respective component were measured by dynamic light scattering and electrophoretic mobility by using a NanoZS (Malvern Instruments, Malvern, UK) operating at a detection angle of 173°.

Thermal and magnetic properties

The thermal stability of the filaments was recorded by measuring the weight change as a function of temperature in N₂ atmosphere using a Perkin Elmer thermogravimetric analysis unit (TGA) 8000 (Perkin, Waltham, MA, USA). The filaments were milled and heated up from 25 °C to 850 °C at a heating rate of 20 °C/min.

The filaments magnetization was measured using a Vibrating Sample Magnetometer (VSM) (DMS, Massachusetts, MA, USA) at room temperature with a magnetic field ranging from -10,000 to 10,000 Oe. The magnetization values per unit mass were adjusted for the magnetic mass content of each sample, determined through TGA analysis.

Mechanical strength

The mechanical strength of filaments was assessed using an Instron 5944 Single Column tabletop Universal Testing System, operating in tensile mode. Prior to testing, all filaments were conditioned at 23 °C in 50% humidity overnight. The used load cell had a capacity

of 100 N, with a gauge length set to 10 mm with an elongation rate of 0.5 mm/min. The filament diameter was measured using a micrometer (Ironsides, graduation 0.001 mm) and at least seven replicates of one sample were measured. The mono magnetic filaments ranged in diameter from approximately 0.07 to 0.15 mm, while the magnetic coaxial filaments from approximately 0.17 to 0.24 mm. The reference mono filaments (without MNPs) measured approximately 0.05 to 0.09 mm in diameter, whereas the reference coaxial filaments measured approximately 0.13 to 0.20 mm.

Surface roughness

The surface topography of the magnetic filaments was evaluated using a 3D Optical Profiler (Sensofar S. Neox) microscope with a white-light vertical scanning interferometry (VSI) mode and CFI60-2 Nikon objective lens of 10× and 50×. The images were processed using the SensoMap software and the following roughness parameters were calculated in multiple areas of each sample: Root mean square roughness (R_q), Total height of profile (R_t), Skewness (R_{sk}), and Kurtosis (R_{ku}) (Gadelmawla et al. 2002).

Wide angle X-ray scattering (WAXS)

WAXS was used to determine the orientation of cellulose and magnetite crystallites in the filaments (Wang et al. 2020). The X-ray intensity data were measured on a Bruker D8 VENTURE PHOTON-III C14 κ-geometry diffractometer system equipped with an Incoatec IμS 3.0 microfocus sealed tube (Cu Kα, $\lambda = 1.54178 \text{ \AA}$) and a multilayer mirror monochromator. Sample diffraction patterns were collected with a 400 mm sample-to-detector distance and with 200 s waiting time. Before evaluation diffraction patterns were corrected by subtracting the background. Based on azimuthal intensity distribution profiles, the orientation index (π) and Herman's orientation parameter (S) were calculated according to Eq. (1) and Eq. (2) (Nishiyama et al. 1997).

$$\pi = \frac{180^\circ - FWHM}{180^\circ} \quad (1)$$

$$S = \frac{3(\cos^2 \gamma) - 1}{2} \quad (2)$$

where FWHM is the full width at the half-maximum (in degrees) of a peak in the azimuthal intensity distribution profile. The average ($\cos^2 \gamma$) is obtained from the average cosine of the azimuthal angle φ based on Eq. (3) and Eq. (4).

$$\langle \cos^2 \gamma \rangle = 1 - 2 \langle \cos^2 \varphi \rangle \quad (3)$$

where,

$$\langle \cos^2 \varphi \rangle = \frac{\sum_{\varphi_0}^{\varphi_0 + \pi/2} I(\varphi) \sin \varphi \cos^2 \varphi}{\sum_{\varphi_0}^{\varphi_0 + \pi/2} I(\varphi) \sin \varphi} \quad (4)$$

where $I(\varphi)$ is the intensity detected at azimuthal angle φ and φ_0 is the azimuthal angle in the beginning of the range used for the calculation of the average cosine ($\langle \cos^2 \varphi \rangle$). S was calculated at a φ_0 of 0, $\pi/2$, π , and $3\pi/2$, and the average of these values is reported.

Results and discussion

Nanoparticle morphology

The TEM images shown in Fig. 1, illustrate the structure of TOCNF and of Fe_3O_4 NPs synthesized by the different methods and loaded in TOCNF (20%) and SA (3.6%). Figure 1a, shows the web-like like structure of TOCNF with a width size distribution of nanofibrils from 5 to 30 nm.

Figure 1b, shows magnetic dispersion (T20/NP25/SA3.6 NH₄OH) with 25% Fe_3O_4 NPs prepared *in-situ*, in the presence of TOCNF (20%) by using NH₄OH as reducing agent. The NPs had a diameter of 7.4 nm, as shown in the respective size distribution histogram. In this magnetic dispersion the NPs are well distinguished within the structure of the nanofibrils.

The TEM images in Fig. 1b to Fig. 1d correspond to the magnetic dispersions (T20/NP25/SA3.6, T20/NP10/SA3.6, and T20/NP5/SA3.6 NaOH) containing 25, 10, and 5% Fe_3O_4 NPs, which were also prepared *in-situ* but using NaOH as a reducing agent. The size histogram of the magnetic dispersions T20/NP25/SA3.6 (NaOH), T20/NP10/SA3.6 (NaOH), and T20/NP5/SA3.6 (NaOH) presented an average diameter size of 7.7, 7.2, and 4.4 nm, respectively.

The TEM images of the magnetic dispersions (NP25/SA3.6) with the NPs synthesized without

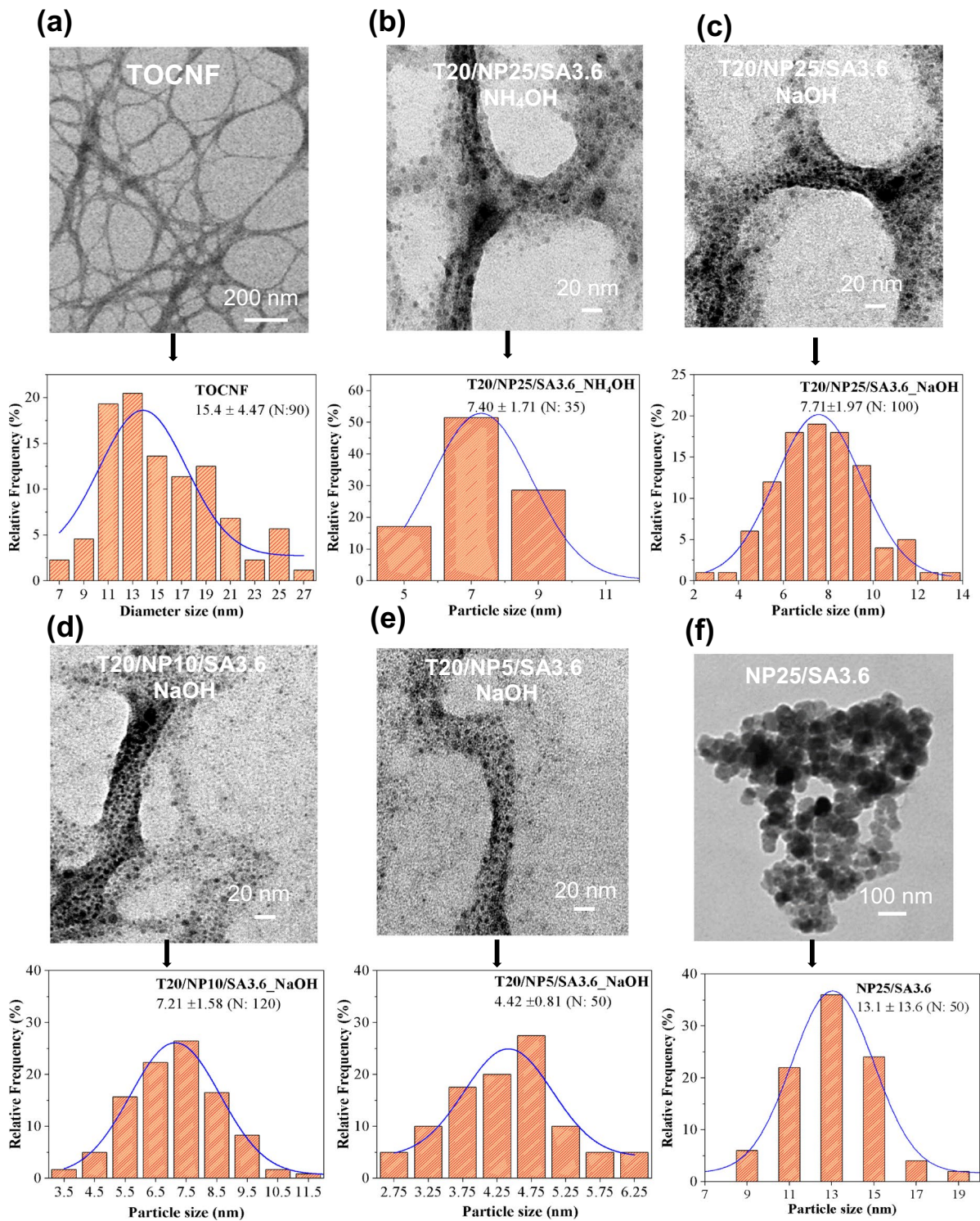


Fig. 1 TEM images of TOCNF **a** and of the magnetic dispersions: T20/NP25/SA3.6_NH₄OH **b**, T20/NP25/SA3.6_NaOH **c**, T20/NP10/SA3.6_NaOH **d**, T20/NP5/SA3.6_NaOH **e**, and

NP25/SA3.6 **f** (with respective size particle distribution histogram below the TEM images)

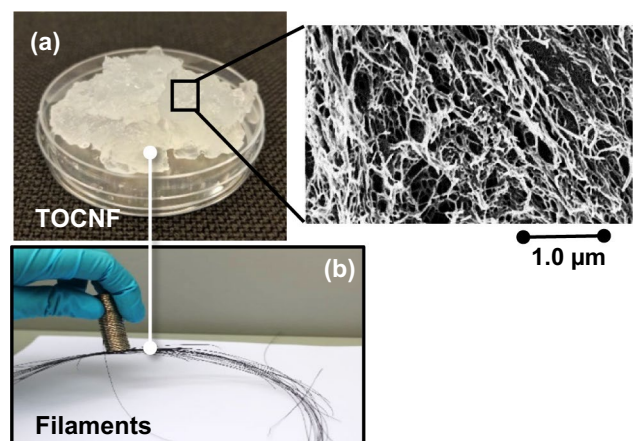
TOCNF (*ex-situ*) are shown in Fig. 1f. The synthesized NPs showed a size spanning from 9 to 20 nm, and presented a high level of aggregation. Clearly, no differences in the NPs size were observed regarding the reducing agent used. However, the most remarkable difference is that the presence of TOCNF, used in the *in-situ* synthesis, stabilized and allowed better control of NPs size and aggregation. NPs prepared *in-situ*, in the presence of TOCNF, revealed an average diameter of 7 nm while *ex-situ* NPs, presented an average diameter size of about 13 nm, as shown in the size distribution histogram of the magnetic solution NP25/SA3.6. This corroborates that the presence of carboxyl groups in TOCNF stabilize NPs (Uddin et al. 2014). However, although NPs prepared in the presence of TOCNF were smaller, individual NPs remained as part of the supernatant, indicating that not all NPs were integrated with the nanofibers.

Filament characterization

Filament composition and morphology

Mono and coaxial filaments were spun using different MNP concentrations synthesized using the *in-situ* (in the presence of TOCNF) and *ex-situ* (absence of TOCNF). The motivation behind the work was to obtain a hybrid inorganic–organic material with unique properties arising from both components. TOCNF presents a gel-like texture and is composed of a matrix of interconnected nanofibrils, as shown in Fig. 2a. The filaments containing TOCNF nanofibrils were dense and rigid, as shown in Fig. 2b.

Fig. 2 A photograph of TOCNF **a** and of the prepared magnetic filament CF/INP25/NH **b**



SEM images in Fig. 3a and Fig. 4a illustrate the surface morphology and respective cross-section at break of the filaments carrying Fe_3O_4 NPs prepared *in-situ* with TOCNF (MF/INP25/NH, CF/INP25/NH, MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA) and *ex-situ* without TOCNF (MF/ENP25 and CF/ENP25), respectively.

All the filaments containing TOCNF showed a relatively rough surface regardless of the MNP preparation method. In contrast MF/ENP25 (Fig. 3a), in the absence of TOCNF, presented a very smooth surface. The cross-section SEM images revealed that all filaments had a circular cross section, but some deformation was observed due to the filaments cryo-cutting in liquid N_2 (Fig. 3a and 4a).

The EDX mapping images and elemental analysis of the filaments cross-section revealed the presence of iron (Fe) (green color) indicating the presence of Fe_3O_4 NPs and carbon (C) element (red color) indicating the presence of either SA and nanocellulose. Iron was heterogeneously dispersed in the mono filaments, MF/ENP25, MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA. However, the three latter filaments displayed a clear reduction of Fe accordingly to the NP concentrations. Regarding the coaxial filaments (CF/INP25/NH and CF/ENP25) an apparent iron core and carbon shell were observed, indicating that TOCNF was uniformly arranged around the magnetic core.

We next measured the surface roughness of the magnetic filaments. Areas of about 0.8 mm^2 from different locations of the samples were scanned, and height variations were recorded. The height and roughness of the profile were determined by ISO

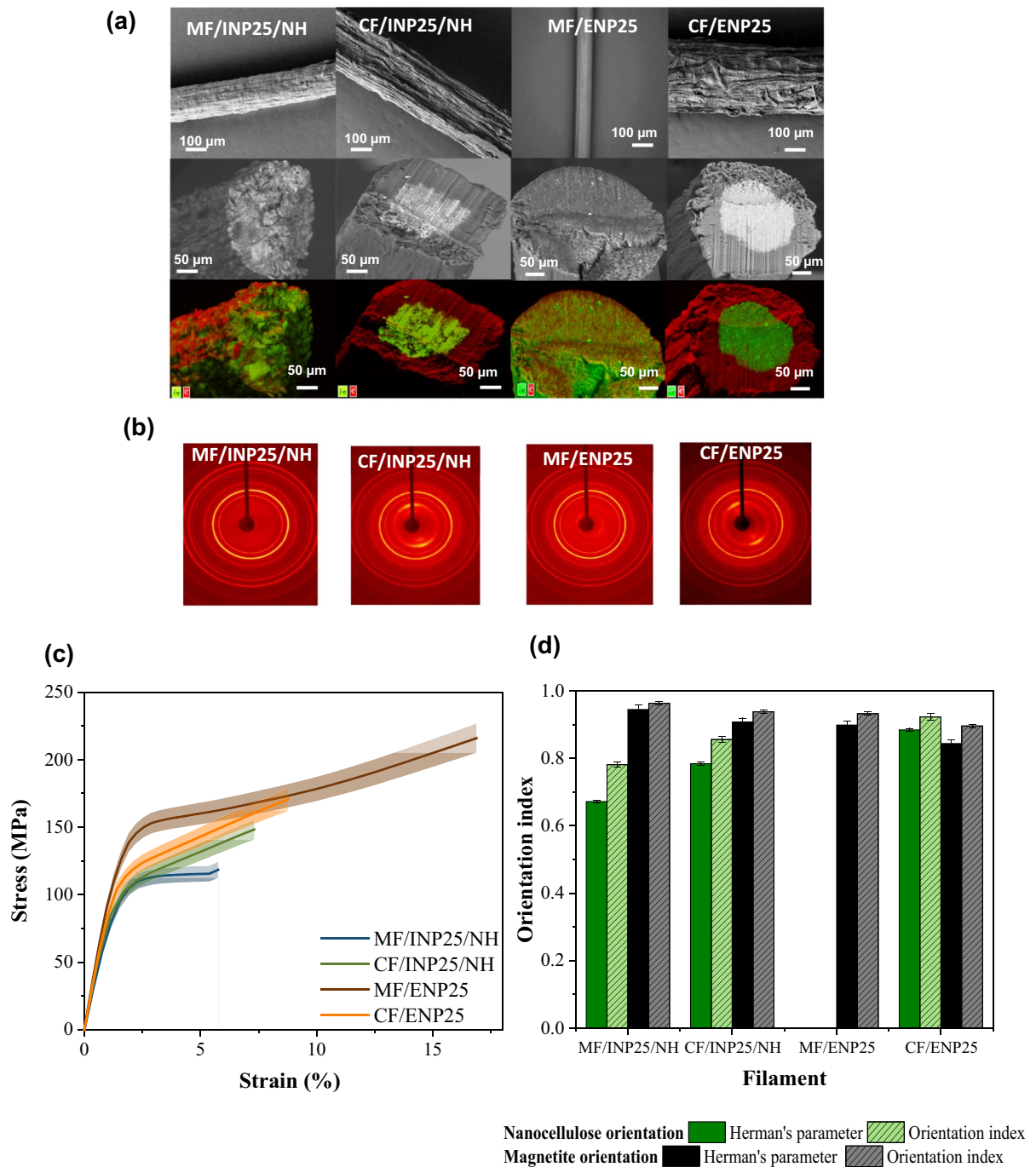


Fig. 3 Main mechanical and structural characteristics of the magnetic filaments MF/INP25/NH, CF/INP25/NH, MF/ENP25, and CF/ENP25: **a** SEM images from cross-sections, **b** Two-dimensional wide-angle X-ray diffraction (2D-WAXS)

images, **c** stress and strain curves (the standard deviation is shown as highlighted areas around the different profiles and **d** Nanocellulose fibril and magnetite nanoparticles orientation

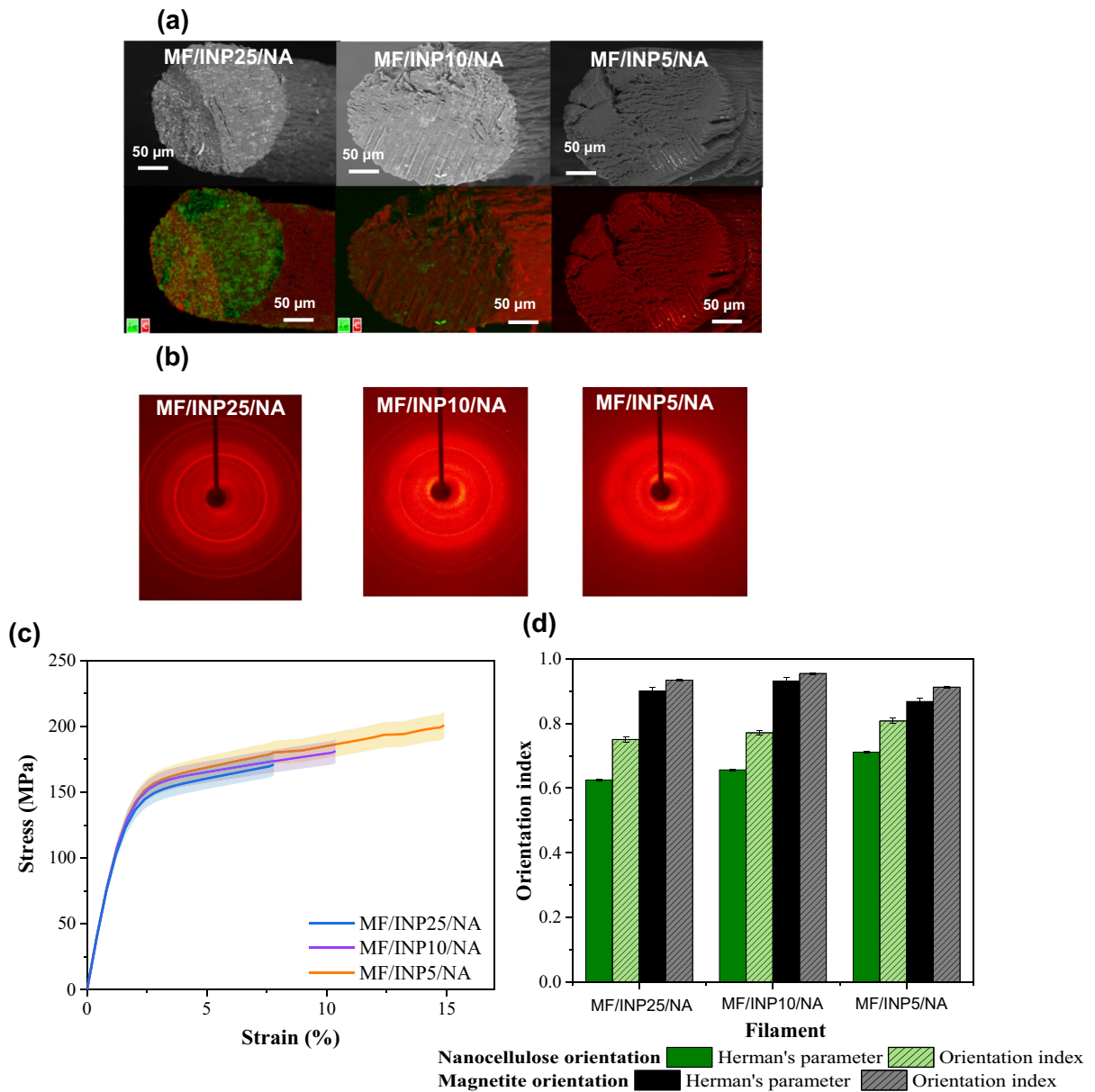


Fig. 4 Main mechanical and structural characteristics of the magnetic filaments MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA: **a** SEM images from cross-sections, **b** Two-dimensional wide-angle X-ray diffraction (2D-WAXS) images, **c**

stress and strain curves (the standard deviation is shown as highlighted areas around the different profiles and **d** cellulose fibril and magnetite nanoparticles orientation

4287:1997, and the calculated parameters are presented in Table S1 and 3D images in Figure S1 (**Supporting information**).

According to the total height (R_t) and roughness (R_q), it is evident that the coaxial filaments (CF/INP25/NH and CF/ENP25) exhibited larger heights

and roughness compared with the mono filaments MF/INP25/NH, MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA. Furthermore, the kurtosis (R_{ku}) parameter revealed that the coaxial filaments (CF/INP25/NH and CF/ENP25) presented a more even and uniform surface ($R_{ku} < 3$), while mono filaments

(MF/INP25/NH to MF/INP5/NA) displayed a sharper surface ($R_{ku} > 3$). Nevertheless, the skewness (R_{sk}) parameter showed a positive value in filaments MF/INP25/NA, MF/INP10/NA, MF/INP5/NA, and CF/ENP25, indicating the presence of more peaks on their surface than valleys, in contrast to filaments MF/INP25/NH, CF/INP25/NH, and MF/ENP25. In the end, filaments in which NPs were prepared *in-situ*, in the presence of TOCNF (MF/INP25/NH, MF/INP25/NA, MF/INP10/NA, MF/INP5/NA, and CF/INP25/NH) exhibited higher roughness compared to filaments MF/ENP25 and CF/ENP25, where the NPs were prepared *ex-situ*, in the absence of TOCNF.

Structural and chemical properties

The magnetic filaments were analyzed by XRD and FTIR to assess their structural and chemical characteristics. Figure 5a shows the FTIR spectra of both TOCNF and magnetic filaments. In the TOCNF spectrum, distinctive peaks at 3300 and 2888 cm^{-1} are attributed to the -OH functional group and -CH₂ stretching vibrations of cellulose (Otenda et al. 2022). Furthermore, the vibrational band at 1604 cm^{-1}

correspond to -C=O group, while the peaks observed at 1044 and 850 cm^{-1} are associated to C–O–C and β -glycosidic linkage vibrations, respectively (Johar et al. 2012; Otenda et al. 2022). Notably, all the characteristic vibrational peaks of nanocellulose were evident in the FTIR spectra of the magnetic filaments.

The peak corresponding to Fe–O, observed at 565 cm^{-1} , overlapped with a wide peak corresponding to TOCNF. In the case of filaments MF/INP10/NA and MF/INP5/NA, the Fe–O peak was negligible due to the low concentration of Fe₃O₄ NPs (10, 5%) compared to TOCNF (20%) and SA (3.6%), respectively.

Figure 5b, shows the X-ray diffraction of both magnetic filaments and TOCNF. In the diffractogram of TOCNF, two distinct peaks appear at $2\theta = 16.0^\circ$ and 22.3° , corresponding to the (1–10/110) and (200) diffraction of cellulose I (Xu et al. 2020; French 2014).

These peaks are also present in the diffractogram of the magnetic filaments, indicating the presence of cellulose crystals with varying intensities. Notably, the main peak at 22.3° is more visible in filaments with a TOCNF shell, while the remaining filaments

Fig. 5 Infrared **a** spectra and X-ray **b** diffractogram of TOCNF and magnetic filaments (red columns in XRD diffractogram represents the diffraction standard pattern of magnetite)

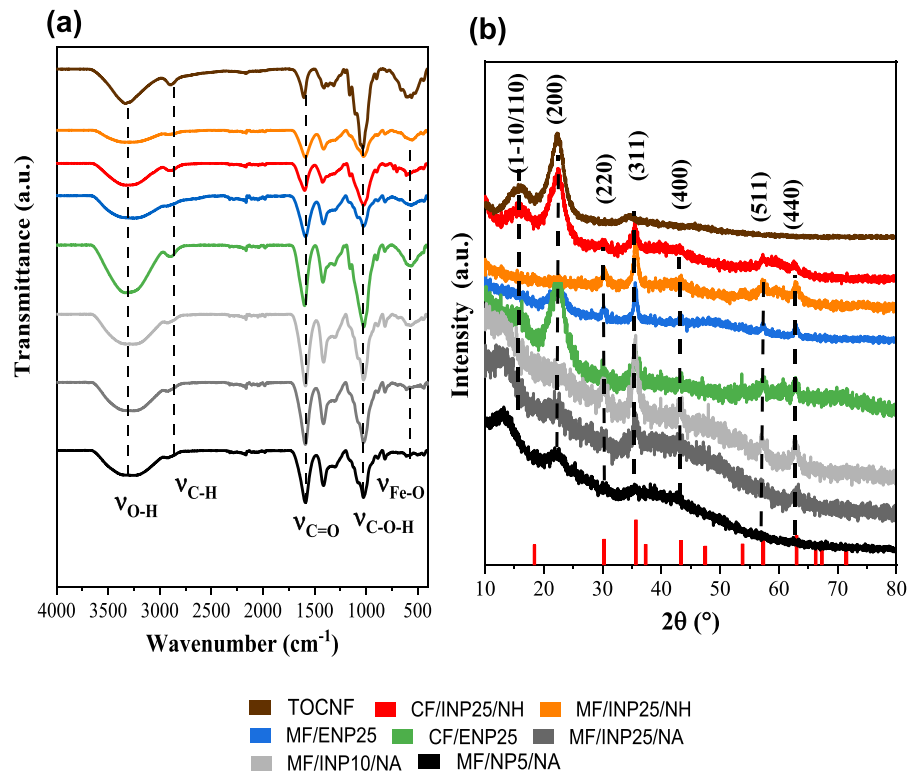


exhibit a broader and less pronounced peak. Additionally, the diffractogram of the magnetic filaments showed diffraction peaks at $2\Theta = 30.1, 35.2, 43.2, 57.3,$ and 62.6° corresponding to the crystallographic planes (220), (311), (400), (511), and (440) of Fe_3O_4 (ICDD: 98–015-8743) (Hastak et al. 2018). The relative intensity of these peaks is correlated with the Fe_3O_4 NP content in the magnetic filaments.

Filaments with NPs prepared in situ, MF/INP25/NH to MF/INP5/NA, showed a decreased intensity and broadening in the main Fe_3O_4 peaks. This suggests that the Fe_3O_4 NPs were small, as observed in TEM, and have a low crystalline quality. This can be attributed to the presence of TOCNF and the coprecipitation procedure.

Mechanical strength and crystallite orientation.

Tensile strength was used to assess the effect of Fe_3O_4 NPs on the mechanical properties of the magnetic filaments. Stress–strain curves of all magnetic filaments are displayed in Fig. 3c and Fig. 4c, respectively. Reference filaments (with no MNPs) revealed a Young's modulus of 10 to 13 (GPa), tensile strength between 210 and 255 (MPa), and a strain at break from 6 to 19%, which are reportedly higher compared to the magnetic filaments (Table S1). Generally, the presence of NPs weakened the mechanical properties of composite materials (Liu and Brinson 2008; Zare 2016).

In this study, at an increased Fe_3O_4 NP content in magnetic filaments from 5 to 25% (MF/INP25/NH, MF/INP25/NA, MF/INP10/NA, MF/INP5/NA), the Young's modulus, tensile strength and strain at break decreased from 14 to 8 GPa, 201 to 118 MPa, and from 15 to 6%, respectively. This effect can be explained by the reduced covalent/hydrogen bonding occurring between NPs and the negative groups of TOCNF and SA (Oprea and Panaitescu 2020). The presence of MNPs at low concentrations enhanced the mechanical properties of the filaments (Nazari 2017).

For instance, the filaments prepared in the absence of TOCNF (MF/ENP25 and CF/ENP25) tolerated a high tensile strength (216 MPa, 171 MPa), strain at break (17%, 9%), and Young's Modulus (10 GPa, 9 GPa), respectively. The addition of TOCNF shell in CF/ENP25 filament increased the strain at break. More studies are needed to explain this behavior and the optimum concentration of Fe_3O_4 NPs to improve the mechanical performance.

WAXS was performed to investigate cellulose nanofiber and Fe_3O_4 orientation in the filaments. The 2D WAXD patterns of the spun filaments are shown in Fig. 3b and Fig. 4b, respectively. Broad and indistinct rings or arcs were observed in the X-ray patterns, suggesting the presence of some crystallinity in the filaments. However, the haziness of the diffraction rings indicated that the degree of crystallinity was relatively low in the filaments MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA. This is likely due to the molecular segment's heterogeneity along the polymer chain and the lower NP concentration.

The orientation index and Herman's parameter are included in Fig. 3d and Fig. 4d. Both parameters are used as an indicator of cellulose and magnetite crystallite orientation, with 0 and 1 corresponding to complete disorder and fully aligned crystals, respectively. Accordingly, filaments with 25% NPs prepared *in-situ* in TOCNF (MF/INP25/NH and CF/INP25/NH) presented a better alignment of magnetite crystals, as indicated by both the orientation index (0.96 and 0.94) and Herman's parameter (0.95 and 0.91), respectively. However, in filaments MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA, with Fe_3O_4 NP content from 25 to 5% and synthesized with the other method, presented less aligned NPs. The latter filament presented the lowest Herman's parameter (0.87) and orientation index (0.91). The filaments, MF/ENP25 and CF/ENP25, with NPs (25%) prepared *ex-situ*, presented an orientation index of 0.93 and 0.90 and Herman's parameter of 0.90 and 0.84, respectively. MF/ENP25 filament reported the best alignment of magnetite crystals after MF/INP25/NH and CF/INP25/NH filaments.

A higher cellulose crystal alignment was observed in filaments (CF/INP25/NH and CF/ENP25) with TOCNF shell reporting an orientation index of 0.86 and 0.92 and Herman's parameter of 0.78 and 0.88, respectively. Clearly, when the cellulose crystals are aligned in these filaments, the (200) diffraction rings are reduced to arcs and become better defined, indicating better fibril orientation. This is further confirmed by the azimuthal intensity profiles (Fig. S3) where the peaks are sharper. The presence of a TOCNF shell increased the fibril alignment in CF/ENP25 compared to the CF/INP25/NH filament. Moreover, a slight improvement in cellulose fibril orientation was observed in filament MF/INP5/NA (NPs content of 5%), reaching an orientation index

of 0.71 and Herman's parameter or 0.81, compared to MF/INP25/NH, MF/INP25/NA, and MF/INP10/NA which reported an orientation index of 0.75, 0.75, and 0.77 and a Herman's parameter of 0.64, 0.63, and 0.66, respectively.

The magnetite crystal orientation and Young's modulus were correlated, since the filaments presenting a higher Young's modulus had a low crystal orientation. Filaments MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA had a Young's modulus of 9, 11, and 14 GPa, respectively. Meanwhile filaments MF/INP25/NH, CF/INP25/NH, MF/ENP25, and CF/ENP25 presented Young's modulus of 8, 9, 10, and 9 GPa.

Thermal stability and magnetic behavior

TGA measurements were performed to investigate the weight loss, thermal stability and degradation of Fe_3O_4 , TOCNF, and SA in the magnetic filaments. Figure 6a shows the TGA curves of all magnetic filaments, which exhibited a three-stage decomposition. The first one (from 100–200 °C) corresponds to elimination of adsorbed water and other volatile components. The second weight loss, between 200 to 450 °C approximately, is due to polymer decomposition of cellulose (around 315 to 400 °C) and SA (around 212 to 426 °C). The third weight loss (around 450 to 800 °C) occurs the degradation

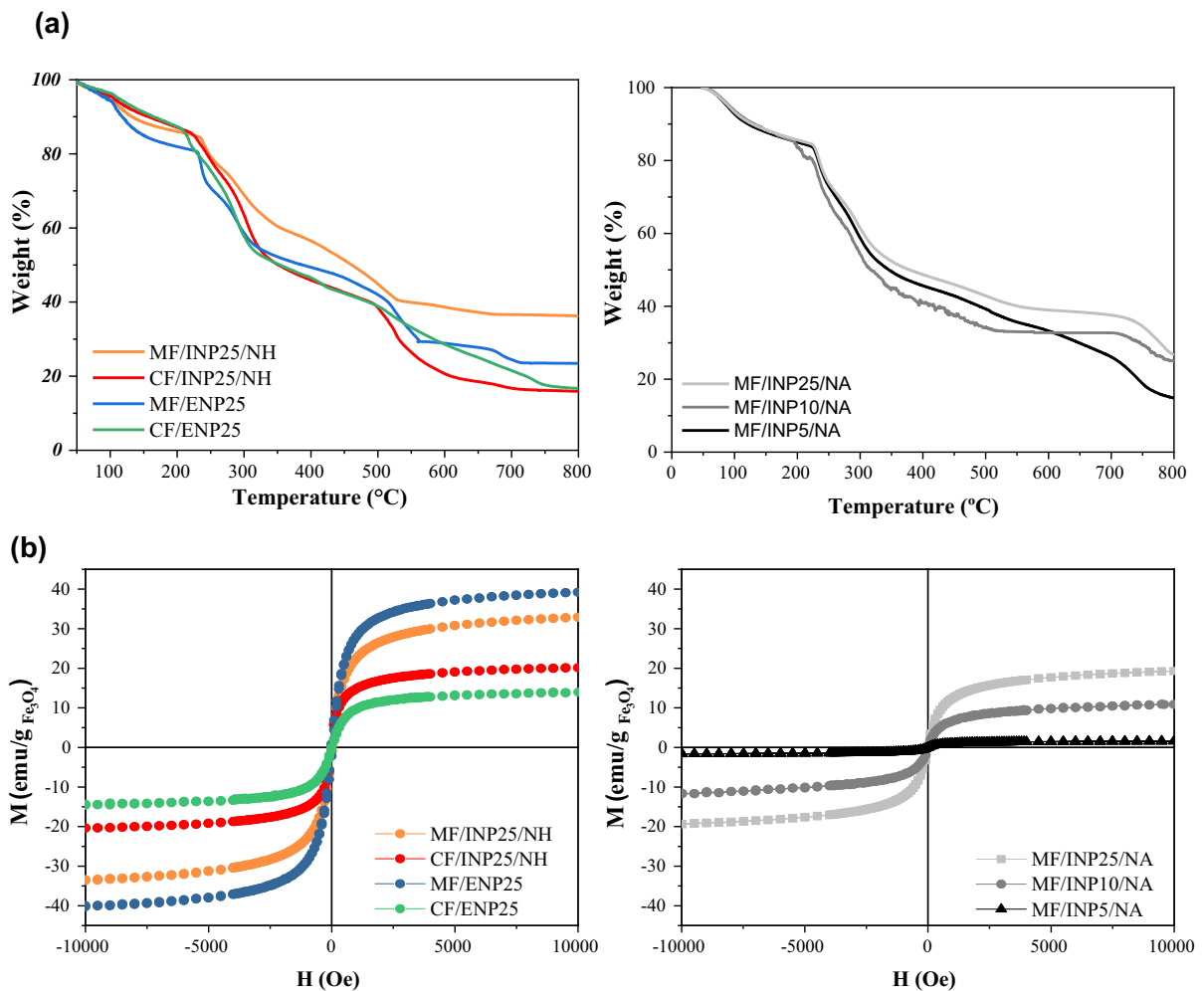


Fig. 6 Thermal stability **a** and magnetization curves **b** of magnetic filaments MF/INP25/NH, MF/ENP25, MF/INP25/NA, MF/INP10/NA, MF/INP5/NA, CF/INP25/NH, and CF/ENP25, respectively

of carbon-based residues. The solid residue that remained above 750 °C corresponds to magnetite, and the amount of magnetic solid content is taken to normalize the magnetization results. This three-stage decomposition is clearly visible in the DTG curves shown in **Fig. S3**, with maximum degradation for the mono filaments occurring around 250 °C and for the coaxial filaments around 315 °C, due to the presence of both biopolymers.

The magnetic properties of the filaments were measured at room temperature using a vibrating sample magnetometer (VSM), with magnetic fields ranging between -10kOe to +10kOe. The specific magnetization values per unit magnetic mass and their corresponding magnetic solid content (%) are reported in Table S1, and magnetization curves, shown in **Fig. 6b**. The role of the synthesis procedure is noteworthy. In this study, the coprecipitation method, the *in-situ* preparation of MNPs in the presence of TOCNF using different reducing agents, resulted in Fe₃O₄ NPs with reduced size and low crystallinity, impacting the magnetic properties. **Figure 6b**, shows the magnetization curves and a reduced magnetization was observed in Fe₃O₄ NPs bonded to TOCNF and SA. This phenomenon aligns with previous research, attributing the magnetization reduction to the encapsulation of MNPs, either by SA or nanocellulose (Denizot et al. 1999; Bedê et al. 2017; Elrhman 2020). Another feature observed is the negligible remnant and coercive force values of the filaments in which the NPs were prepared *in-situ*, in the presence of TOCNF, confirming their superparamagnetic behavior. Moreover, filaments MF/INP25/NA, MF/INP10/NA, and MF/INP5/NA containing Fe₃O₄ NPs prepared in situ (25, 15, and 5%), showed a reduced magnetization, that agrees with the decreased crystalline size and quality of the MNPs, as confirmed by XRD, TEM, and WAXS analyses. The filaments MF/INP25/NH and MF/INP25/NA, even though had the same ratio of Fe₃O₄ NPs (25%), showed differing magnetic properties due to the use of different precipitating agents for MNP synthesis. This difference can be attributed not only to the NPs size but also to differences in crystalline quality, which are influenced by the reactants used (Yazid and Joon 2019; Ba-Abbad et al. 2022). Understanding this complex magnetic behavior lies beyond the scope of

this study and additional research is planned in the near future.

Conclusion

Mono and coaxial magnetic filaments were prepared by wet spinning dispersions of TOCNF (20%), SA (3.6%), and MNPs (5–25%). MNPs were synthesized *in-situ* with TOCNF using NH₄OH and NaOH, and *ex-situ* without TOCNF, followed by SA addition. Fe₃O₄ NPs and TOCNF significantly affected the filaments' properties. NPs synthesized with TOCNF were smaller (5–9 nm) and less aggregated. Filament MF/INP25/NH (25% NPs, NH₄OH reducing agent) showed higher magnetization and better crystal alignment than MF/INP25/NA (25% NPs, NaOH reducing agent), while the latter had better mechanical properties. Meanwhile the mono filament without TOCNF, exhibited higher mechanical strength and magnetization but lower crystal alignment and the coaxial filaments (CF/INP25/NH and CF/ENP25) overcame poor orientation and mechanical strength issues. Filaments with TOCNF had a more porous and rougher surface, and high MNP loading reduced density and crystallinity, impacting mechanical strength and thermal stability. These hybrid materials have potential applications in micro-sensors, artificial muscles, flexible microwave absorption textiles, and data storage media, guided by the identified cause-effect relationships.

Author contributions Conceptualization: L.d.C.A., Y.P.R., M. B.; Methodology: L.W.; Formal analysis and investigation: L.d.C.A, L. W., M.A.G.G., P.G.A., A.A.P.; Writing-original draft preparation: L.d.C.A.; Writing-review and editing: L.W, Y.P.R., O.J.R., J.R.; Funding acquisition, Resources, Supervision: O.J.R, J.R.

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Data availability No datasets were generated or analysed during the current study.

Declarations

Conflict of interest The authors declare no competing interests.

Ethical approval. Ethics approval was not required for this research.

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