

N₂O emissions from a single-stage partial nitrification/anammox granule-based reactor – a model based assessment

Martin Morset*, Borja Valverde-Pérez^{*,1}, Jan Michael Blum^{*,2}, Carlos Domingo-Félez^{*}, Miguel Mauricio-Iglesias^{**}, Barth F. Smets^{*}

* METlab, Department of Environmental Engineering, Technical University of Denmark Building 115, 2800 Kongens Lyngby, Denmark

** Department of Chemical Engineering, Universidade de Santiago de Compostela, 15782, Santiago de Compostela, Spain

¹ corresponding author: bvape@env.dtu.dk; ² presenting author

1. INTRODUCTION

PN/A based treatment has many benefits:

- Lower aeration costs (63% [1])
 - Lower sludge production (90% [2])
- Certain operational conditions cause N₂O emissions which could offset the carbon footprint of PN/A!

Objectives:

- 1) Develop a model to predict N₂O emissions from granular Anammox reactors
- 2) Define optimal operational conditions for maximum nitrogen removal and minimum N₂O emissions

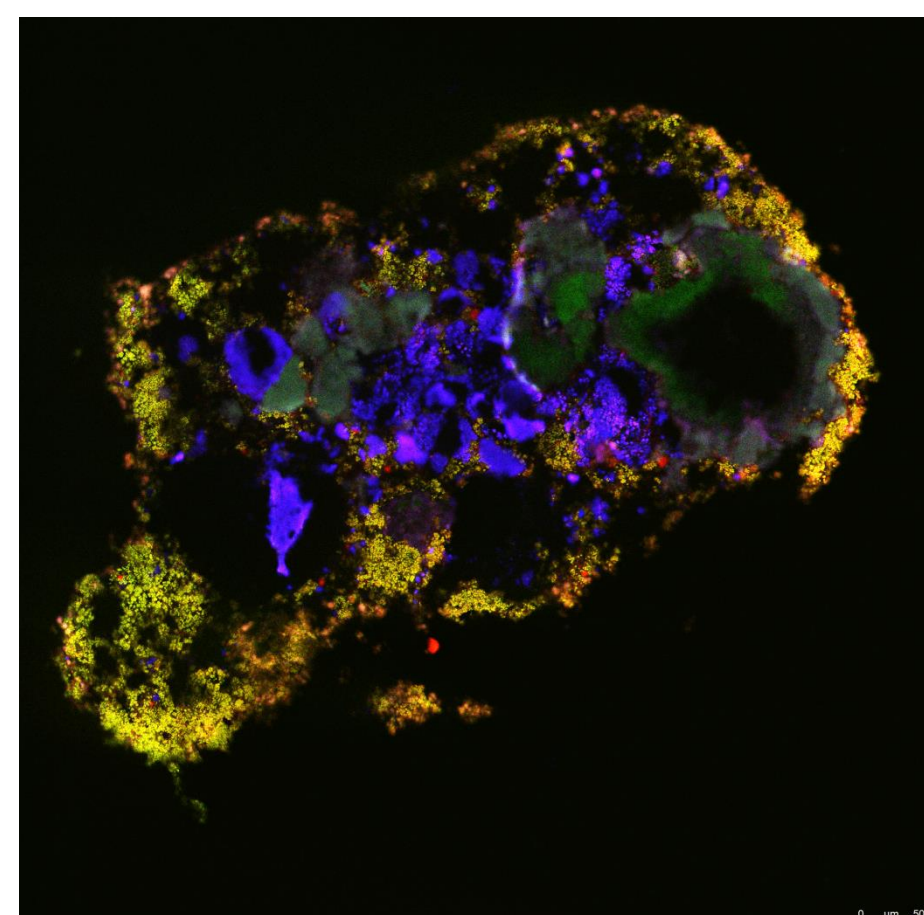
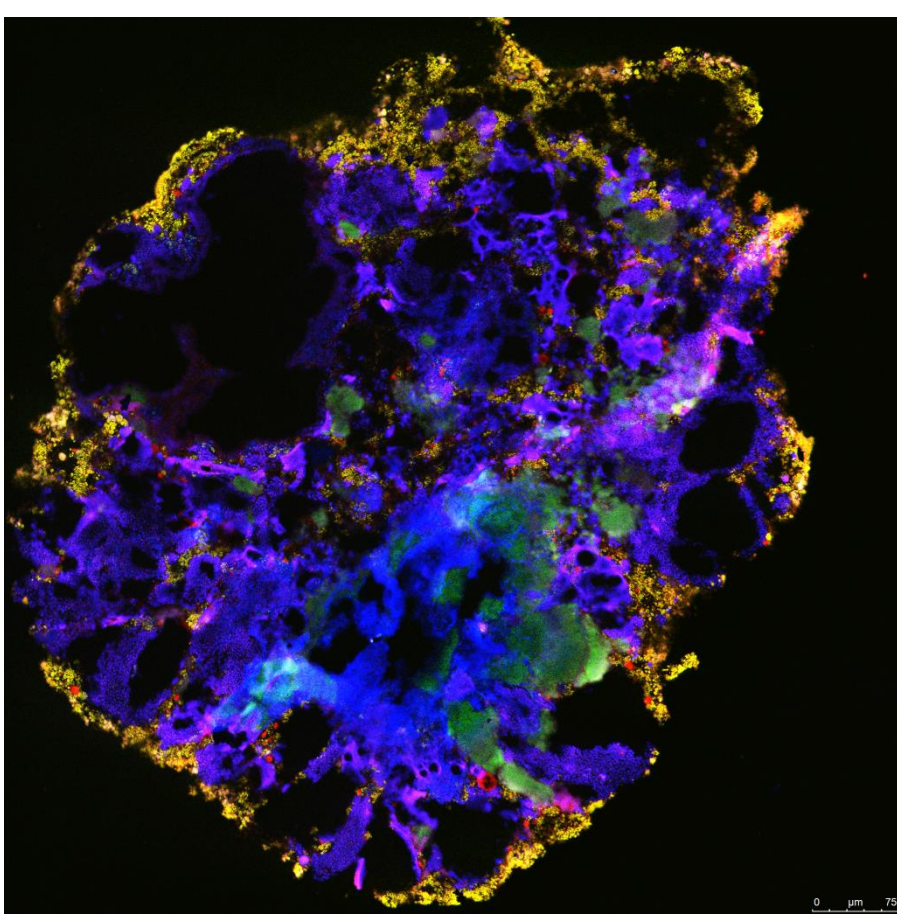
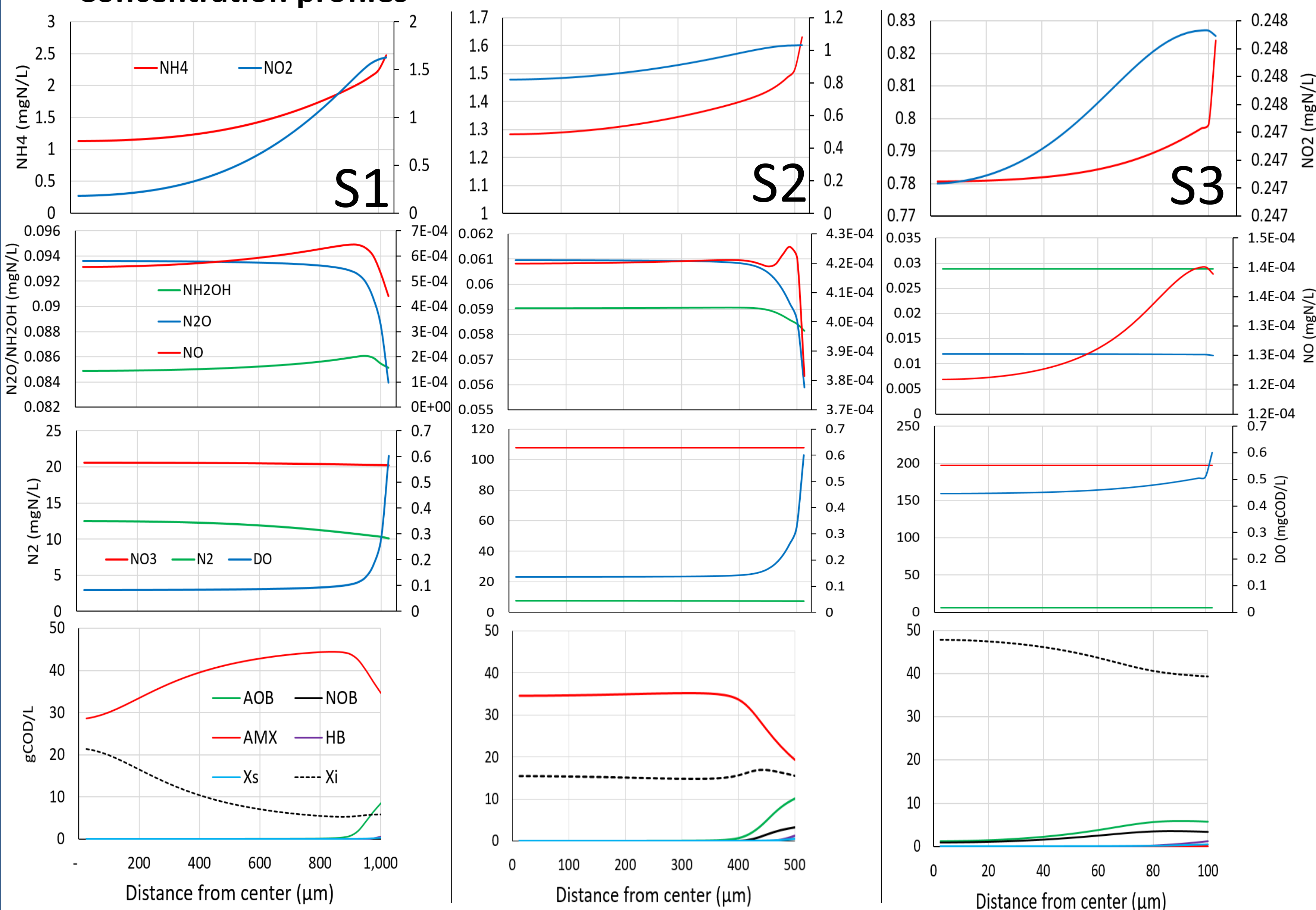
3. SCENARIOS

Process performance analyzed in 4 different scenarios – granule size and alkalinity loading:

Parameter	S1	S2	S3	S4
N-load (mgNH ₄ -N/L-d)	200	200	200	200
DO _{bulk} (mg/L)	0.6	0.6	0.6	0.6
pH _{bulk} (-)	7.50	7.50	7.50	7.50
Radius (μm)	1000	500	100	1000
NH ₄ :HCO ₃ _{in}	1.32	1.32	1.32	0.32

4. RESULTS

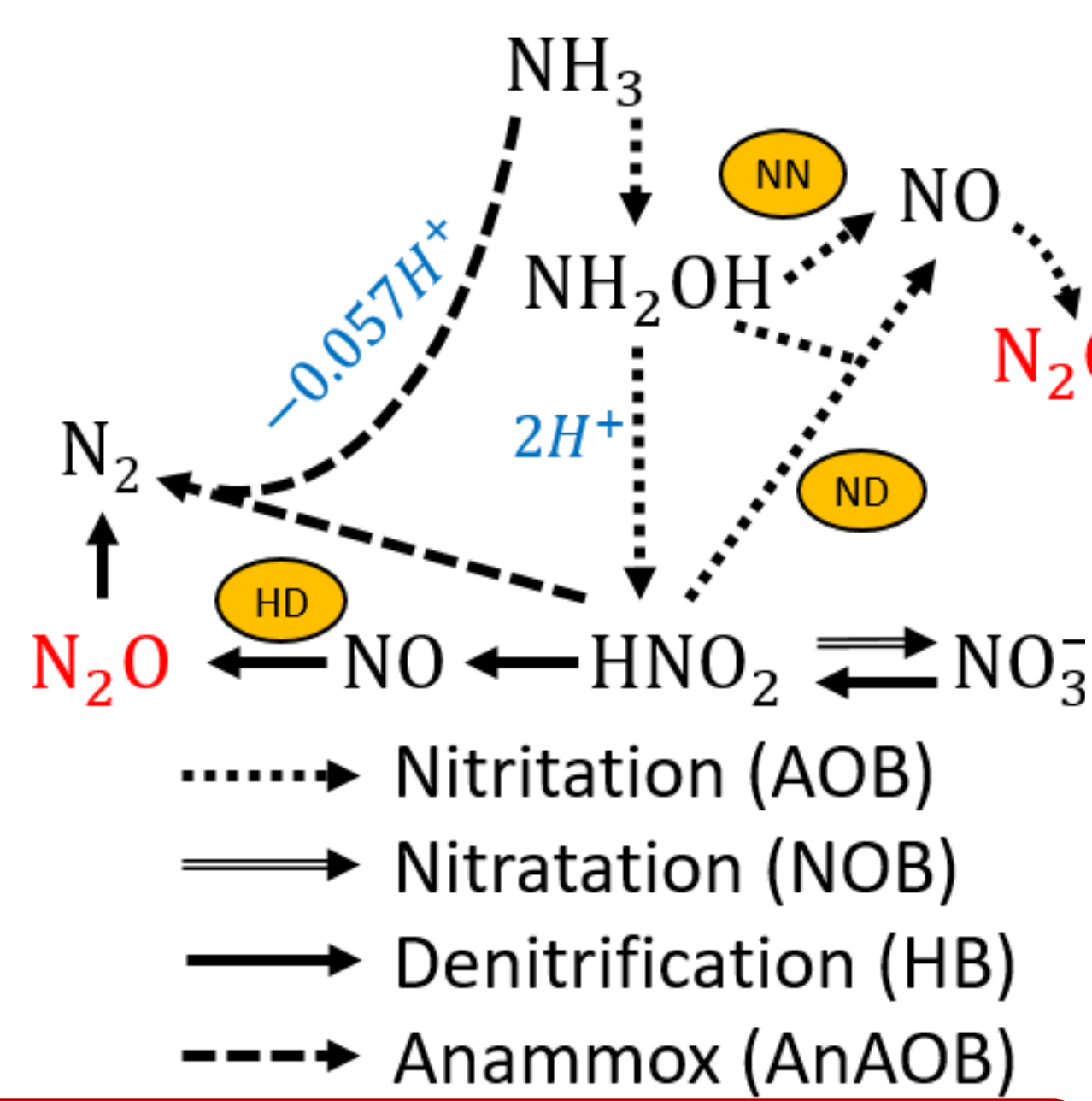
Concentration profiles



Granules in laboratory reactor

2. MODEL DEVELOPMENT

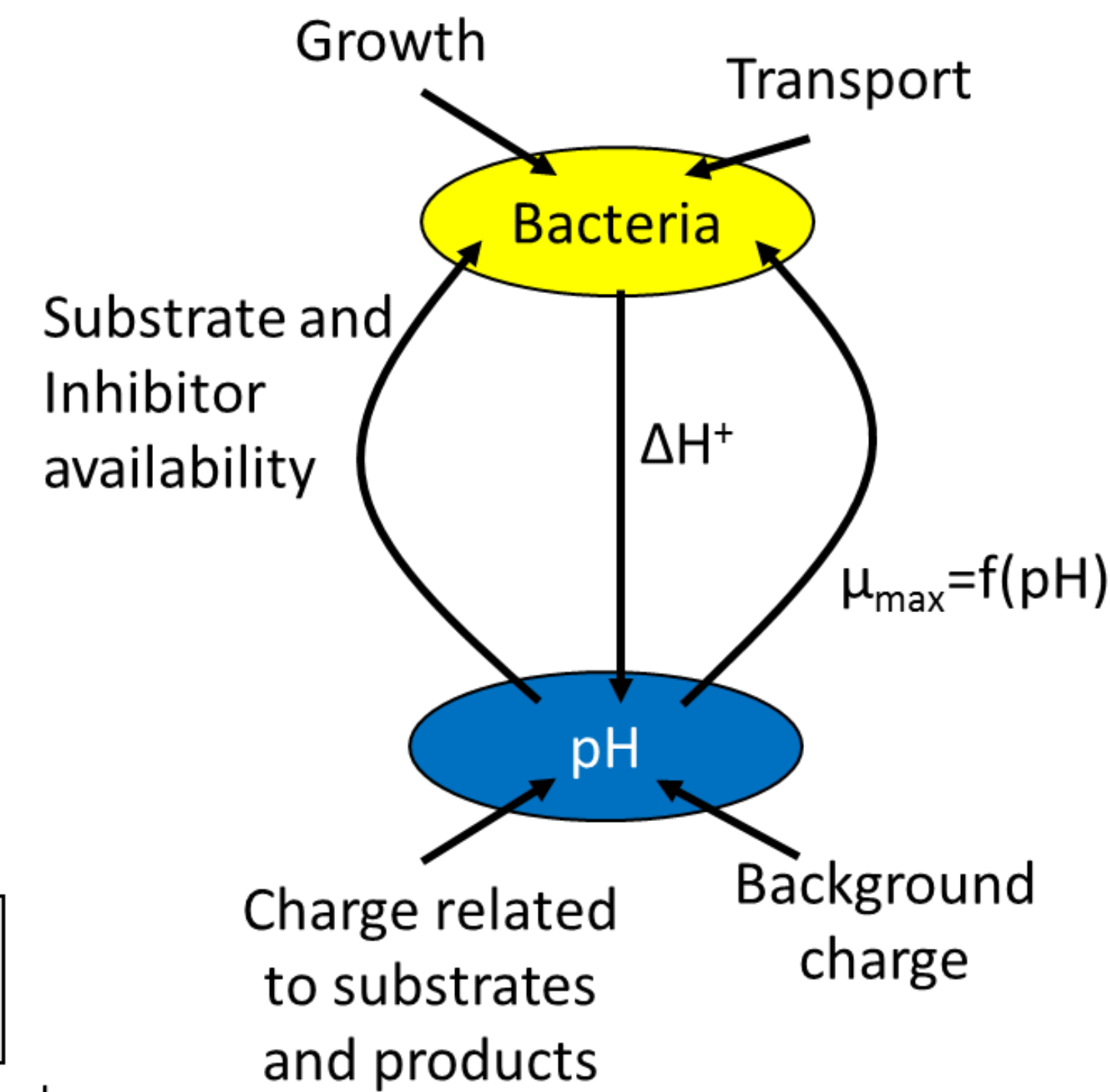
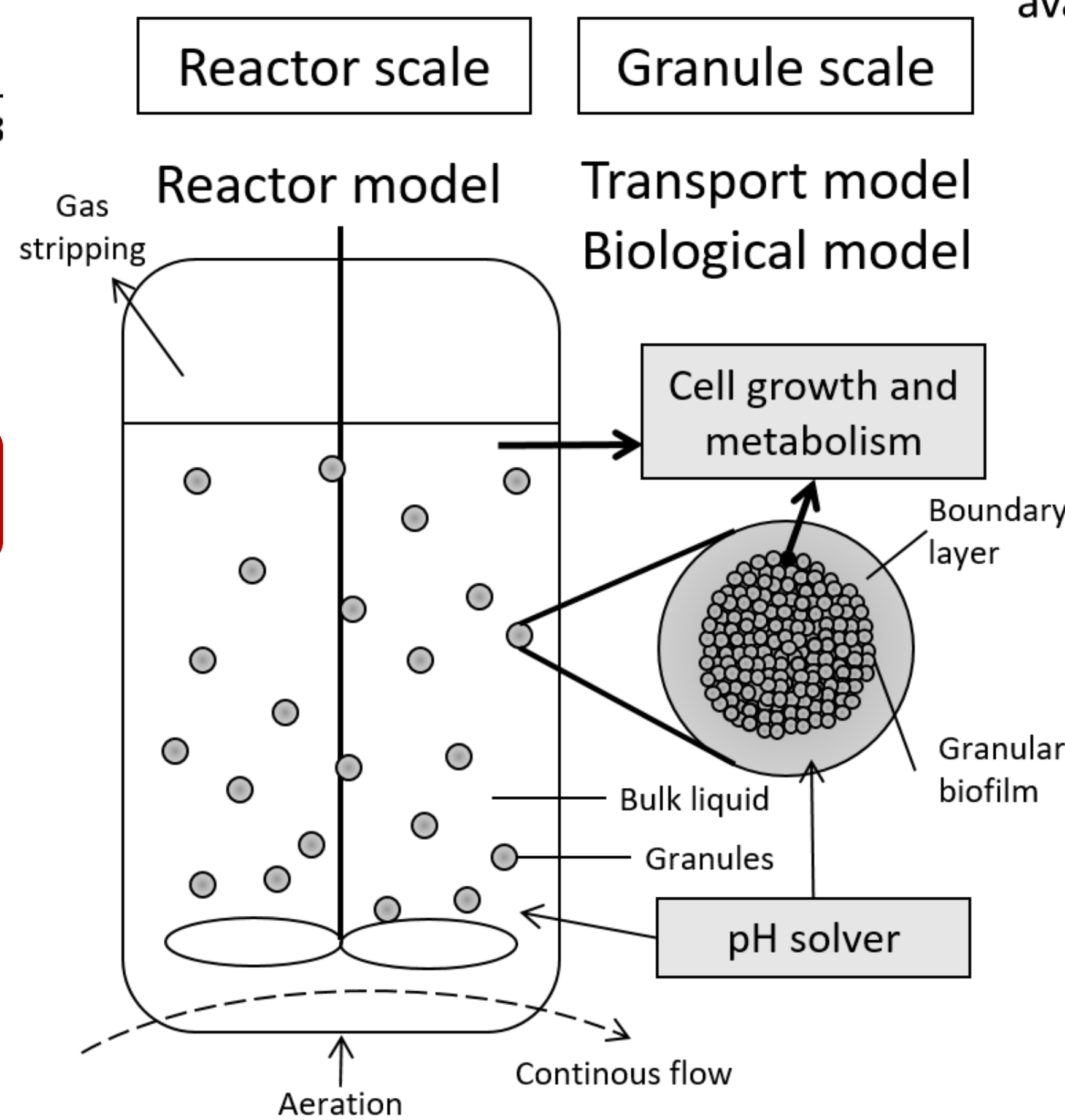
1D model combining NDHA [3], Advection-Diffusion approximation and pH solver



PDE-system discretized in space

Two-scale model:

- CSTR with granular biomass
- Bulk and granule connected via mass transfer resistance (boundary layer)



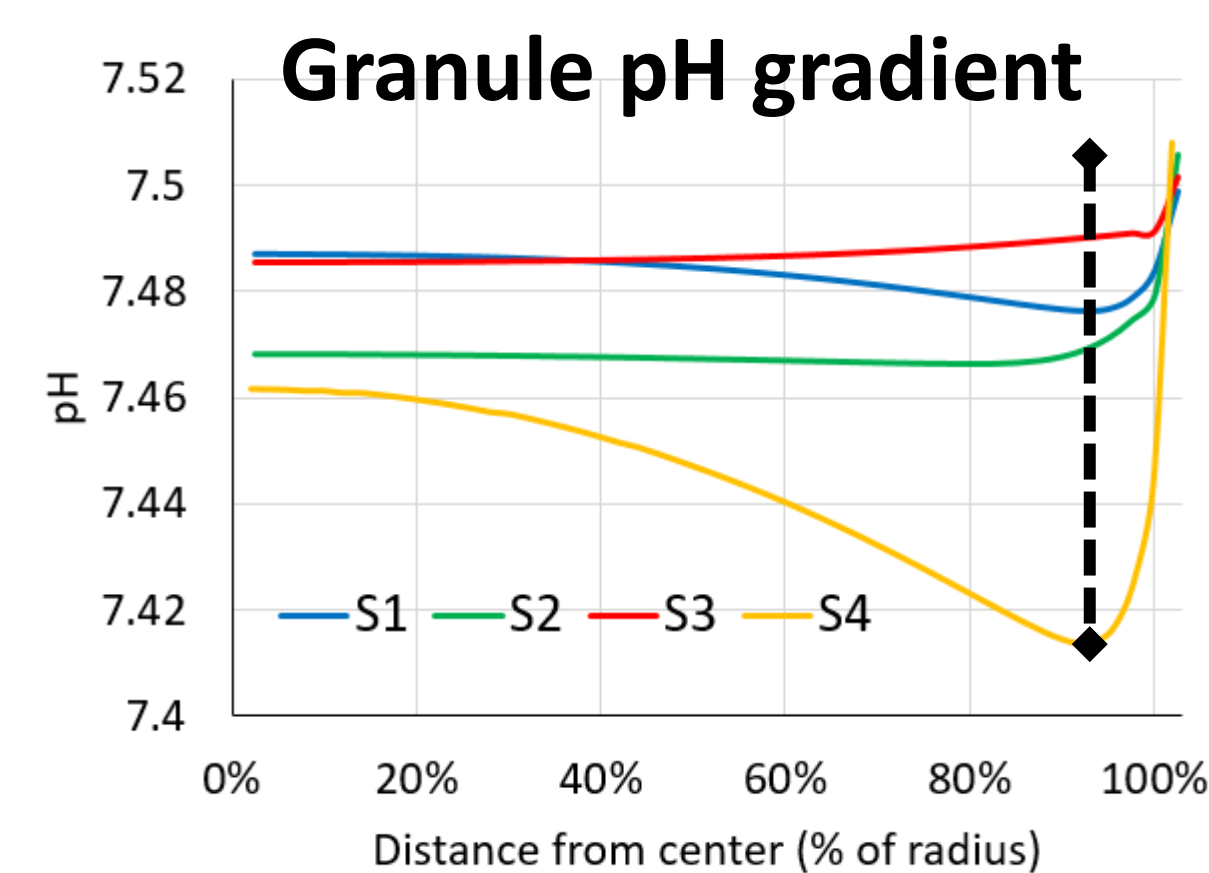
pH modeling:

- Brent-Dekker method
- Guaranteed convergence
- pH affects microbial activity by
 - Impact on enzymatic activity and basic cellular functions
 - Speciation of substrates and inhibitors

Simulation	L _{O2} (mgO ₂ -N/L-d)	N ₂ O _e /ΔTN (%)	ΔTN/TN _{in} (%)
S1	398	2.1	87.8
S2	678	4.8	44.7
S3	972	0.6 (a)	0.0
S4	398	3.4	87.2

(a) Reported as N₂O_e/ΔNH₄

- Large granules (S1) support Anammox growth and lead to NOB suppression.
- Higher N₂O emissions in small granules (S2). NOB grow and N-removal is halved compared to S1.
- At high oxygen load (S3) Anammox are outcompeted and all ammonia is nitrified.
- In S4, alkalinity limitations caused increased N₂O production through the ND pathway due to more HNO₂ accumulation compared to S1.



- Excess buffer capacity and H⁺ consumption by Anammox [4] counteracts H⁺ production by AOB in granule at alkaline conditions [5] [6]

ΔpH < 0.1

- pH gradient is steep where there is net H⁺ production (AOB)

ACKNOWLEDGEMENTS

The work has been funded by The Danish Council for Independent Research Technology and Production Sciences (FTP) (Project N2Oman, File No. 1335-00100B).

N2OMan

References:

- [1] Volcke et al. *Water Science and Technology* 56.9 (2007): 117-125.
- [2] Mulder et al. *FEMS microbiology ecology* 16.3 (1995): 177-183.
- [3] Domingo-Félez et al. *Environmental Science: Water Research & Technology* 2.6 (2016): 923-930.
- [4] Lotti et al. *Water research* 60 (2014): 1-14.
- [5] Vanggaard et al. *Water Science and Technology* 67.11 (2013): 2608-2615.
- [6] Charles et al. *Applied and environmental microbiology* 83.6 (2017): e02985-16.

